

# Materials Science

Dr. K.E.D. Sumanasiri

## Textbooks:

Callister W.D. Materials Science and Engineering, An Introduction, John Wiley & sons (e-book posted in LMS)

Jones D.R.H., Michael F.A. Engineering Materials 1; An Introduction to properties, Applications and design, Butterworth - Heinemann Publishers

What is materials science?

Why should we know about it?

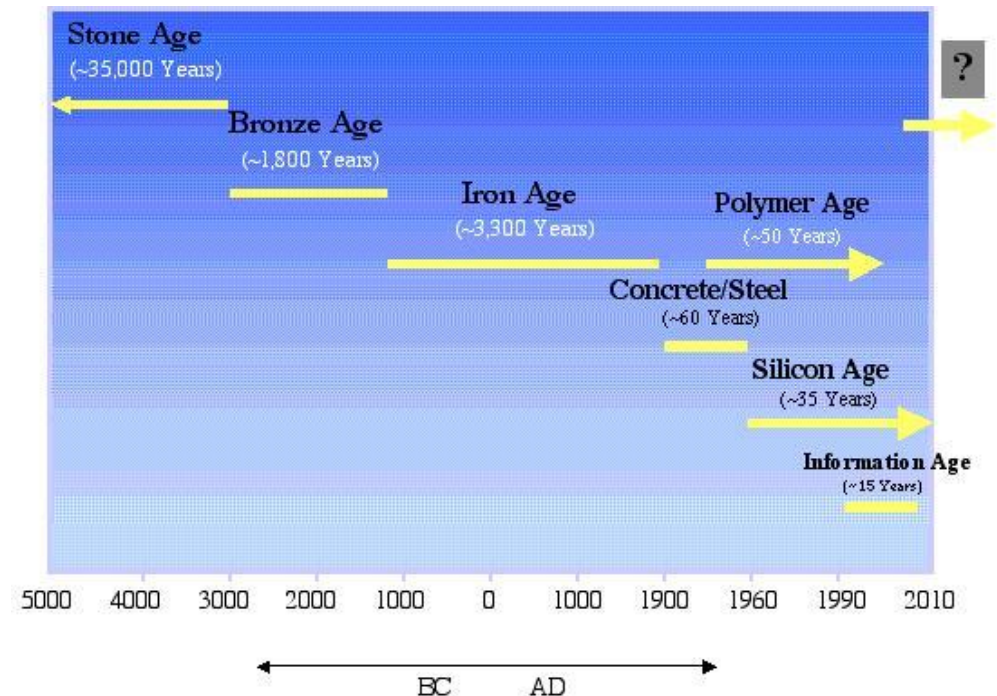
# *A Brief Introduction to Materials Science and Engineering*

# Why the Study of Materials is Important?

- Production and processing of materials constitute a large part of our economy.
- Engineers choose materials to suite design.
- New materials might be needed for some new applications.
- Modification of properties might be needed for some applications.

# History of Materials Science & Engineering

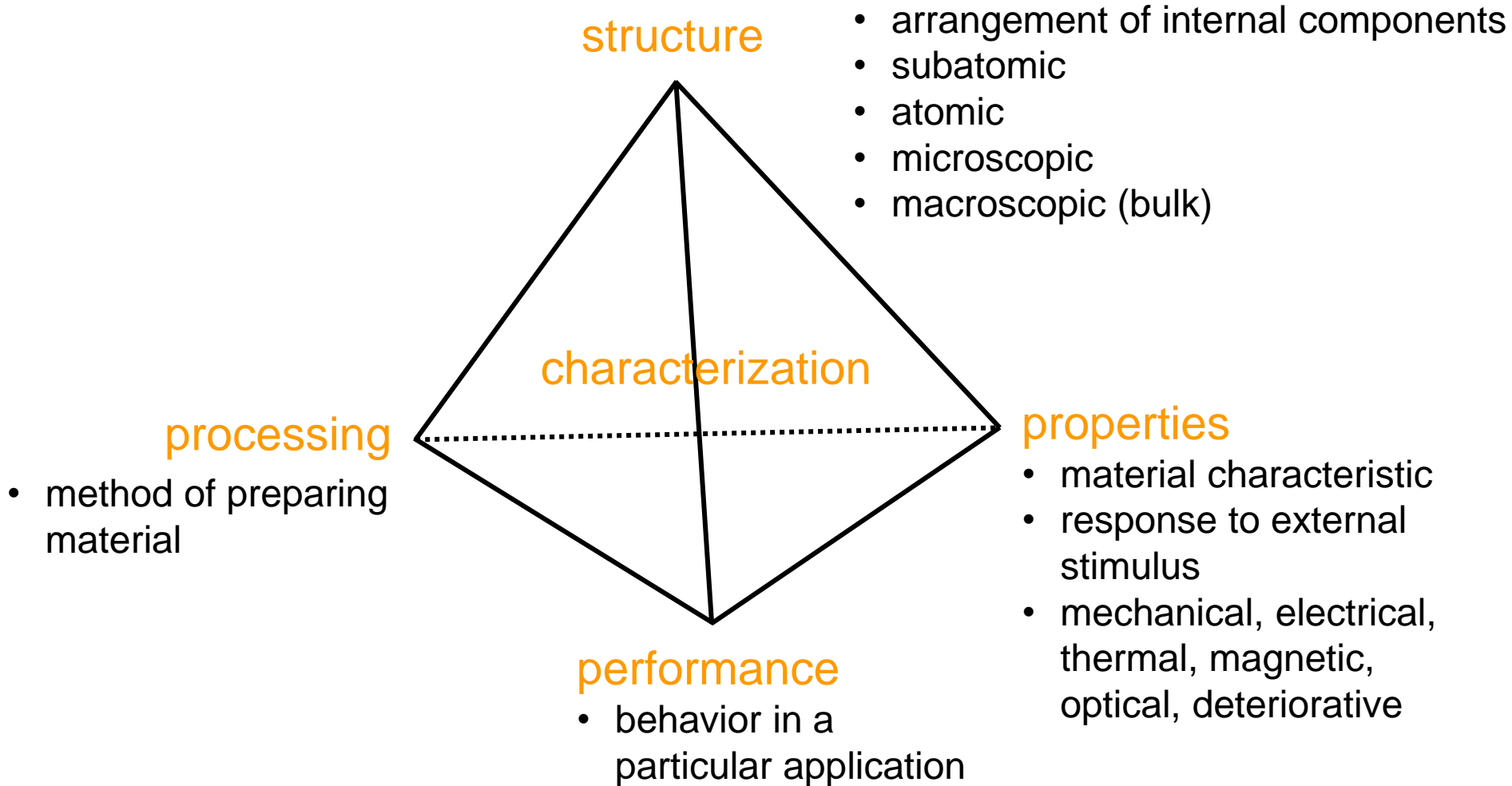
- materials closely connected our culture
- the development and advancement of societies are dependent on the available materials and their use
- early civilizations designated by level of materials development



- initially natural materials
- develop techniques to produce materials with superior qualities (heat treatments and addition of other substances)

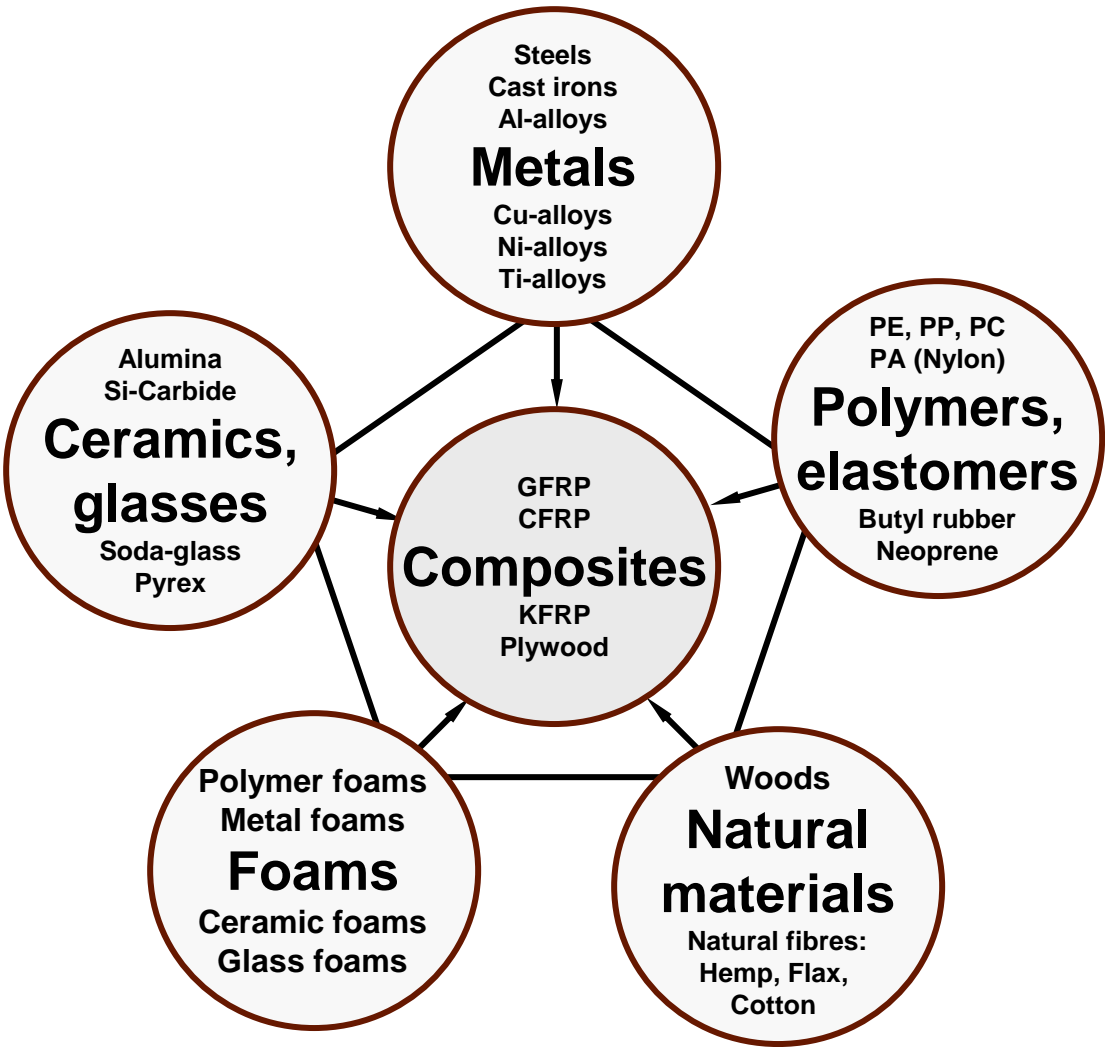
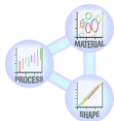
## MATERIALS SELECTION!

# *Materials Science and Engineering*





# The world of materials – Year 2000



# Classification of Materials

## Metals

- good conductors of electricity and heat
- lustrous appearance
- susceptible to corrosion
- strong, but deformable



## Ceramics & Glasses

- thermally and electrically insulating
- resistant to high temperatures and harsh environments
- hard, but brittle



## Polymers

- very large molecules
- low density, low weight
- maybe extremely flexible





# *Classification of Materials: A Few Additional Categories*

## Biomaterials

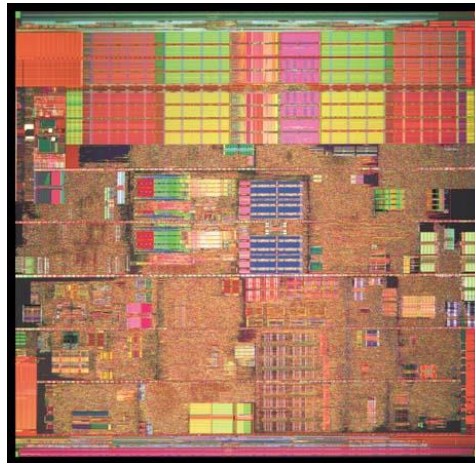
- implanted in human body
- compatible with body tissues



hip replacement

## Semiconductors

- electrical properties between conductors and insulators
- electrical properties can be precisely controlled



Intel Pentium 4

## Composites

- consist of more than one material type
- designed to display a combination of properties of each component



fiberglass surfboards

# Why study materials?

- applied scientists or engineers must make material choices
- materials selection
  - in-service performance
  - deterioration
  - economics

BUT...really, everyone makes material choices!

aluminum



glass



plastic



# Types of Materials

- **Metals:**

- Strong, ductile
- High thermal & electrical conductivity
- Opaque, reflective.



- **Polymers/plastics:** Covalent bonding → sharing of e's

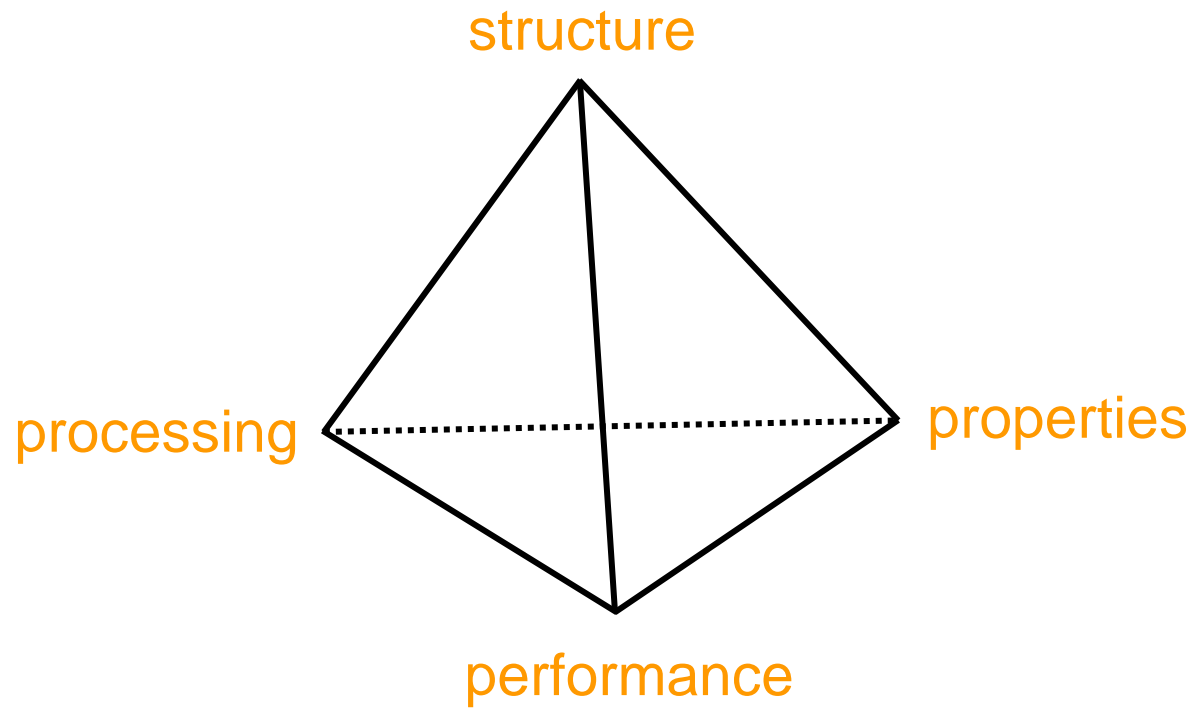
- Soft, ductile, low strength, low density
- Thermal & electrical insulators
- Optically translucent or transparent.



- **Ceramics:** ionic bonding (refractory) – compounds of metallic & non-metallic elements (oxides, carbides, nitrides, sulfides)

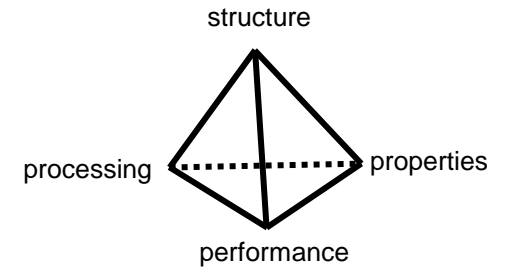
- Brittle, glassy, elastic
- Non-conducting (insulators)





# *Levels of Structure*

Sub-atomic, atomic, microscopic, bulk



**STRUCTURE** (length scale)



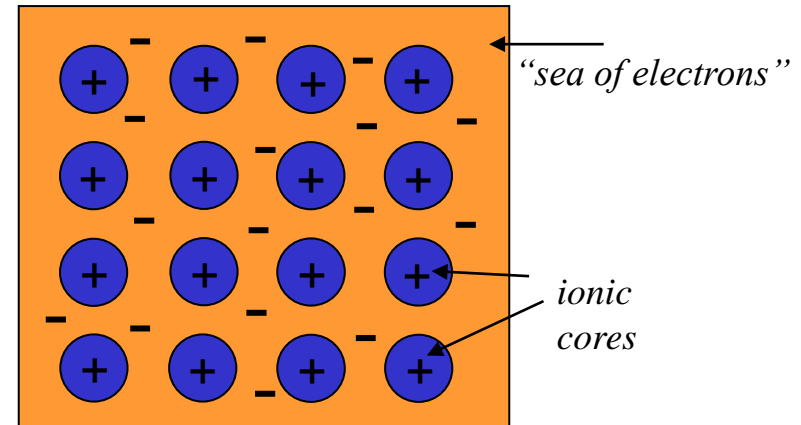
**Sub-atomic**

< 0.2 nm

# Metals

## Metallic Bond

- one, two, or three valence electrons
- valence electrons free to drift through the entire material forming a “*sea of electrons*” surrounding net positive *ionic cores*
- non-directional bond



# Periodic Table of the Elements

GROUP																								VIII								
	IA																															
1	1	H																							2	He						
		IIA																		IIIB		IVB	VB	VIB		VIIA						
2	3	Li	4	Be																	5	B	6	C	7	N	8	O	9	F	10	Ne
3	11	Na	12	Mg																	13	Al	14	Si	15	P	16	S	17	Cl	18	Ar
PERIOD	4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36													
		K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr													
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54														
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe														
6	55	56		72	73	74	75	76	77	78	79	80	81	82	83	84	85	86														
	Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn														
7	87	88																														
	Fr	Ra																														
											</																					

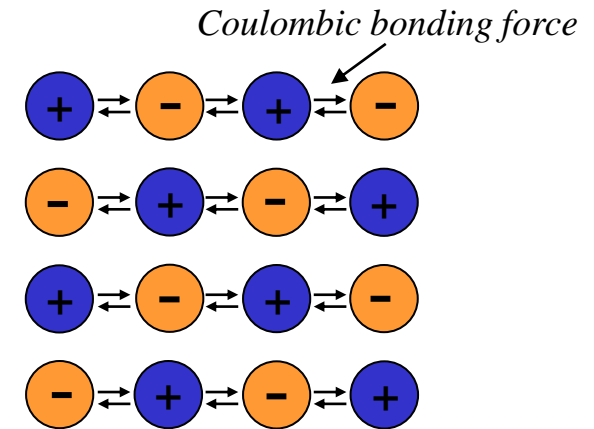
## Properties

- good conductors of electricity and heat
- lustrous appearance
- susceptible to corrosion
- strong, but deformable

## Ceramics and Glasses

# Ionic Bond

- composed of metallic and non-metallic elements
- metallic elements give up valence electrons to non-metallic elements
- all atoms have filled “inert gas” configuration
- ionic solid
- non-directional bond



GROUP		Periodic Table of the Elements																VIII			
IA																		IIB			
1	1	H																	2	He	
2	2	Li	Be																	10	Ne
3	3	Na	Mg																	18	Ar
4	4	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr		
5	5	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe		
6	6	Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn		
7	7	Fr	Ra																		
				57	58	59	60	61	62	63	64	65	66	67	68	69	70	71			
				La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu			
				89	90	91	92	93	94	95	96	97	98	99	100	101	102	103			
				Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr			

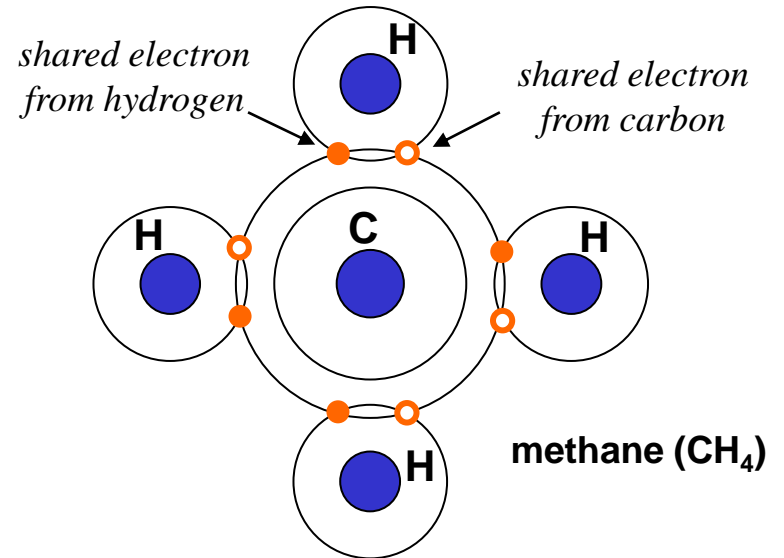
## Ceramics & Glasses

- thermally and electrically insulating
- resistant to high temperatures and harsh environments
- hard, but brittle

# Polymers

# Covalent Bond

- electrons are shared between adjacent atoms, each contributing at least one electron
- shared electrons belong to both atoms
- directional bond



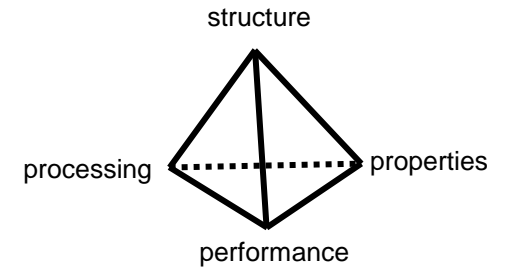
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6	Cs	Ba		72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	81	Tl	82	Pb	83	Bi	84	Po	85	At	86	Rn																
7	Fr	Ra																																															
				57	La	58	Ce	59	Pr	60	Nd	61	Pm	62	Sm	63	Eu	64	Gd	65	Tb	66	Dy	67	Ho	68	Er	69	Tm	70	Yb	71	Lu																
				89	Ac	90	Th	91	Pa	92	U	93	Np	94	Pu	95	Am	96	Cm	97	Bk	98	Cf	99	Es	100	Fm	101	Md	102	No	103	Lr																

# Polymers

- very large molecules
- low density, light weight materials
- maybe extremely flexible



# Levels of Structure

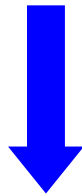


**STRUCTURE** (length scale) →



**Sub-atomic**

< 0.2 nm  
1 nm = ?



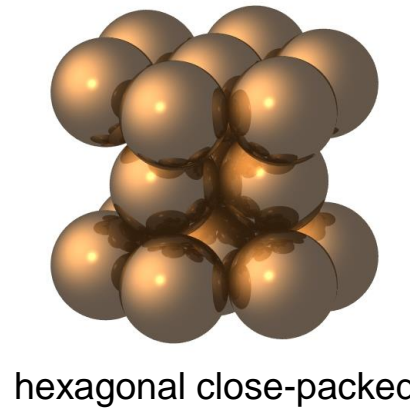
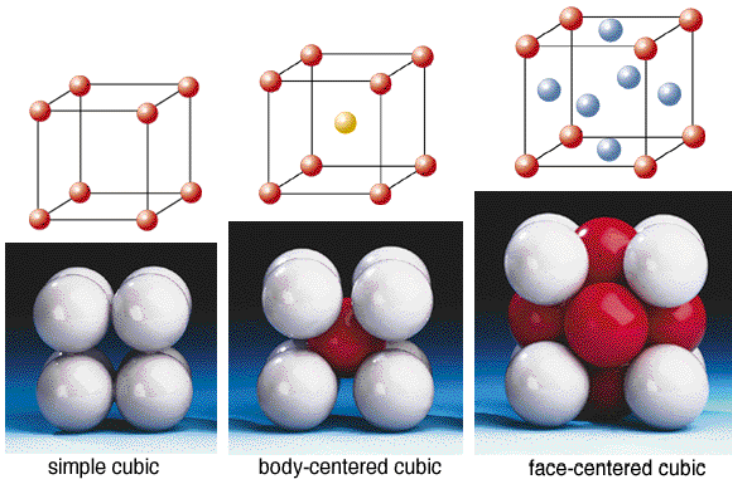
**Atomic**

0.2-10 nm

# Atomic Arrangement: Ordered vs. Disordered

## Crystalline:

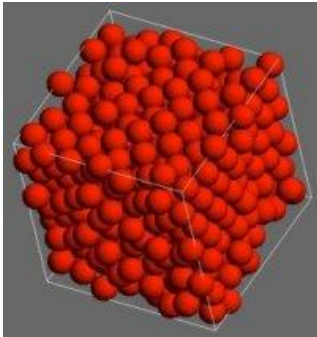
atoms are arranged in a 3D, periodic array giving the material “*long range order*”



- stacking can effect properties (i.e. ductility)
- anisotropic materials

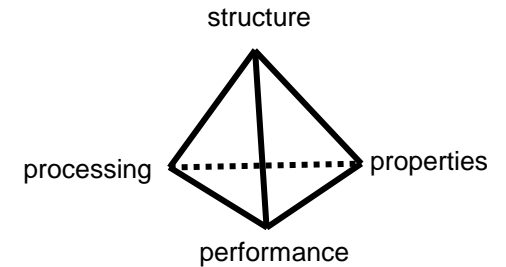
## Non-crystalline or amorphous:

atoms only have short-range, *nearest neighbor order*



- viscous materials (generally complex formulas) or rapid cooling
- isotropic materials

# Levels of Structure

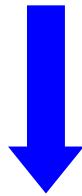


**STRUCTURE** (length scale)



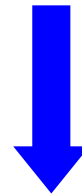
**Sub-atomic**

< 0.2 nm  
1 nm = ?



**Atomic**

0.2-10 nm



**Microscopic**

1-1000  $\mu\text{m}$

# Microstructure

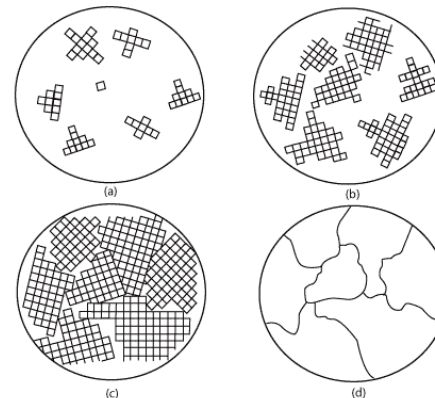
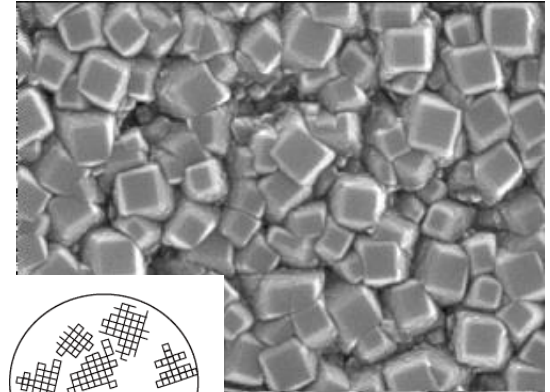
## Single Crystal

- the periodic arrangement of atoms extends throughout the entire sample
- difficult to grow, environment must be tightly controlled
- anisotropic materials

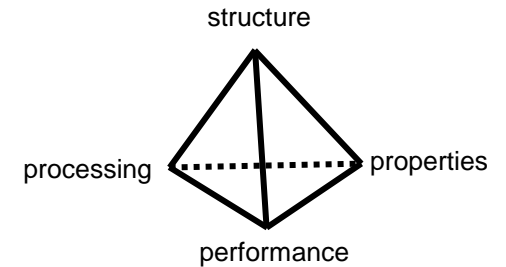


## Polycrystalline

- many small crystals or grains
- small crystals misoriented with respect to one another
- several crystals are initiated and grow towards each other
- anisotropic or isotropic materials



# Levels of Structure

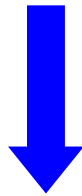


**STRUCTURE** (length scale)



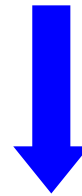
**Sub-atomic**

< 0.2 nm  
1 nm = ?



**Atomic**

0.2-10 nm



**Microscopic**

1-1000  $\mu\text{m}$



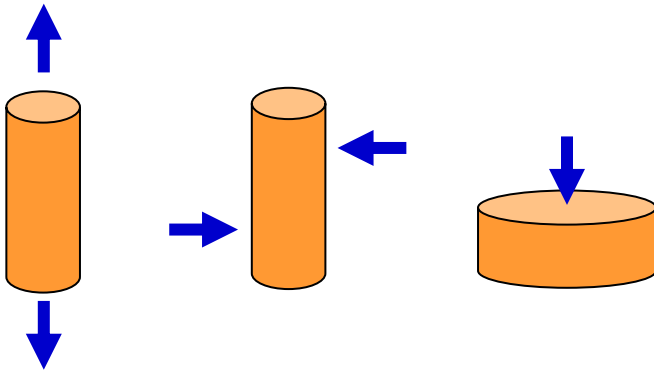
**Bulk**

> 1 mm

# Bulk Properties

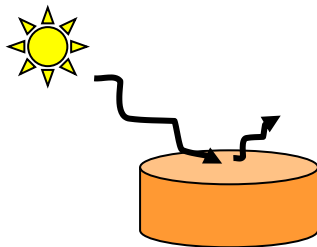
## Mechanical:

elastic modulus  
shear modulus  
hardness



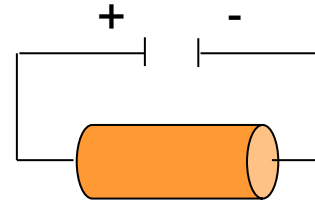
## Optical:

reflectivity  
absorbance  
emission



## Electrical:

conductivity  
resistivity  
capacitance



## Thermal:

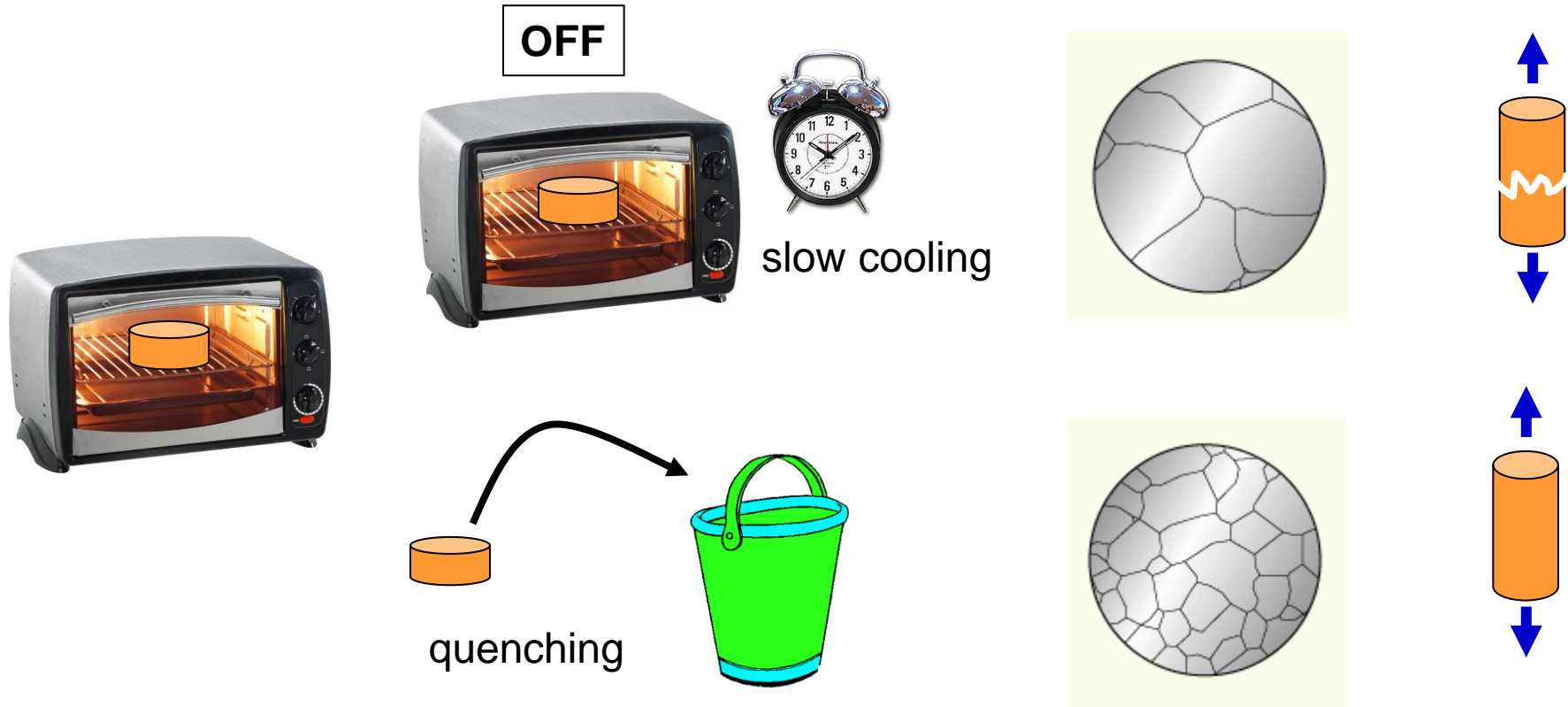
thermal expansion  
heat capacity  
thermal conductivity



# *Processing → Structure → Properties → Performance*

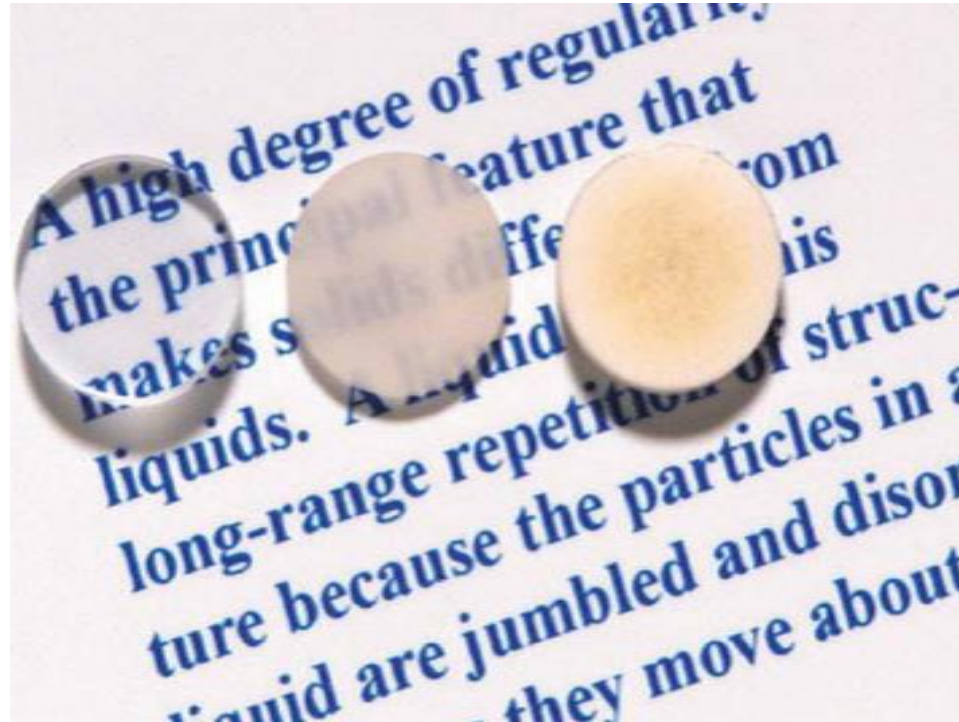
Performance Goal: increased strength from a metallic material

In actuality, crystals are NOT perfect. There are **defects**!  
In metals, **strength** is determined by how easily defects can move!



# *Processing → Structure → Properties → Performance*

## Aluminum Oxide ( $\text{Al}_2\text{O}_3$ )



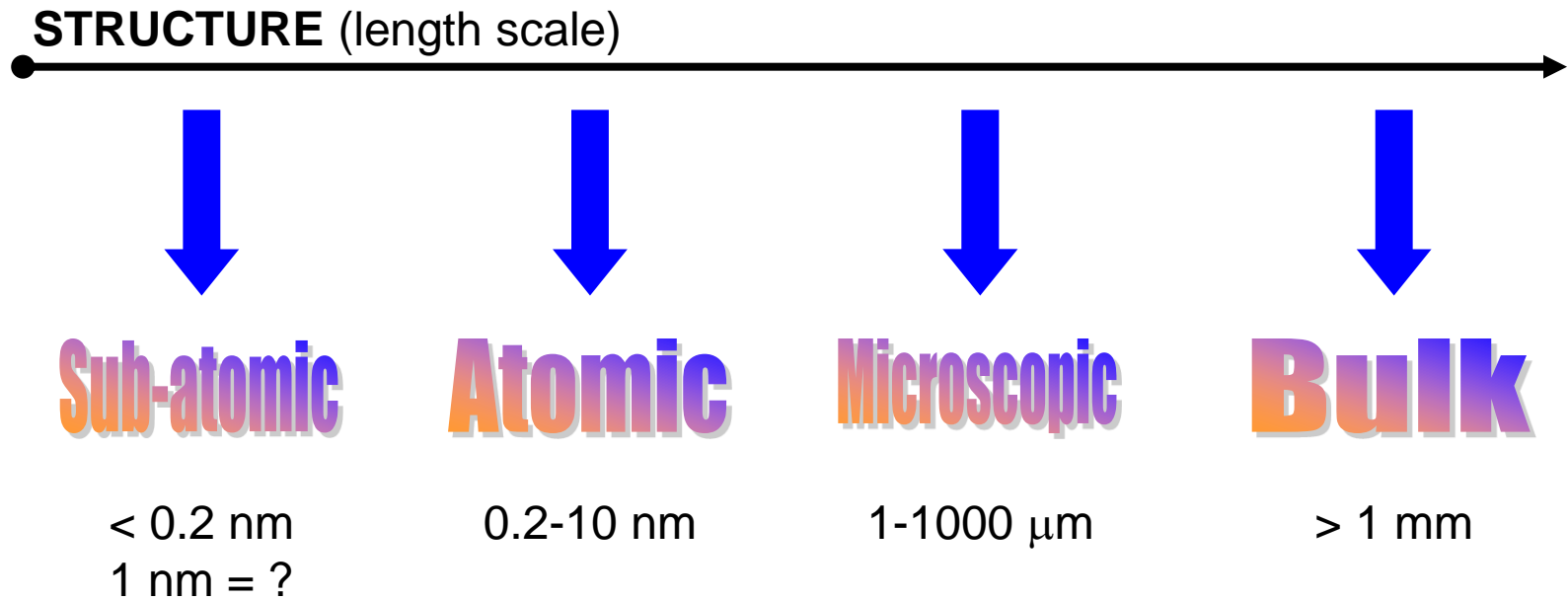
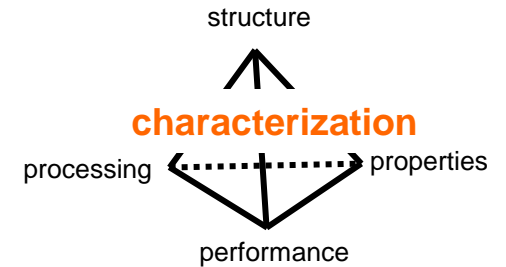
single-crystal  
(transparent)

polycrystalline,  
fully dense  
(translucent)

polycrystalline,  
5% porosity  
(opaque)

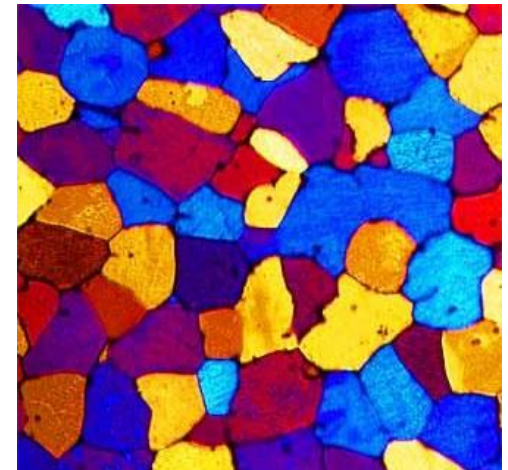
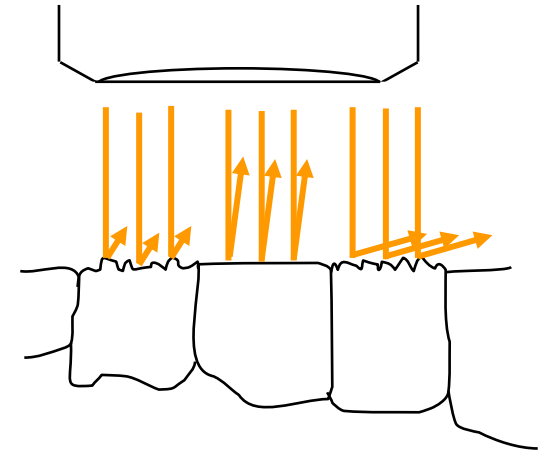
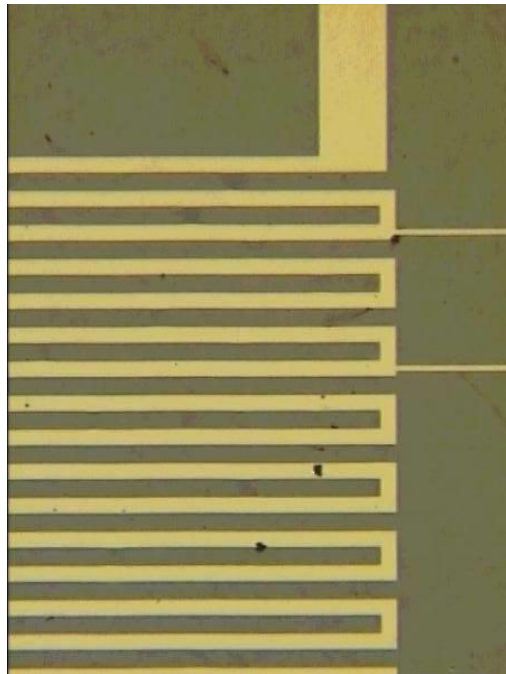
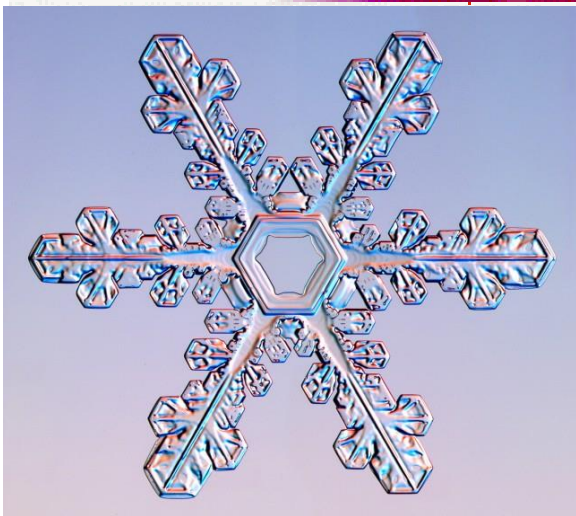
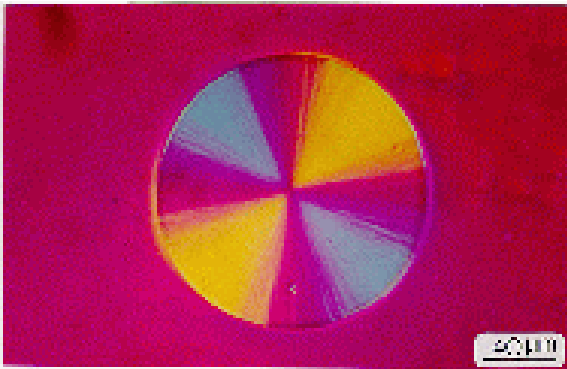


# Characterization Techniques



# Optical Microscopy

- light is used to study the microstructure
- opaque materials use reflected light, where as transparent materials can use reflected or transmitted light

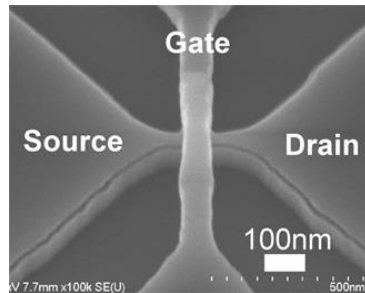
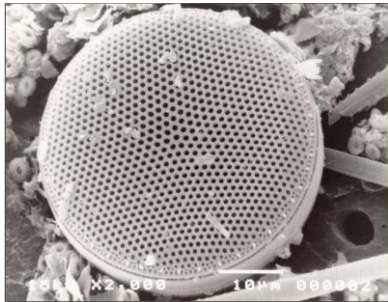


# Electron Microscopy

- beams of electrons are used for imaging
- electrons are accelerated across large voltages
- a high velocity electron has a wavelength of about 0.003 nm
- the electron beam is focused and images are formed using magnetic lenses
- reflection and transmission imaging are both possible

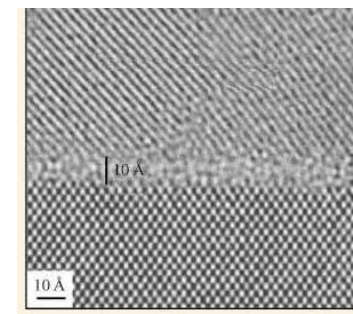
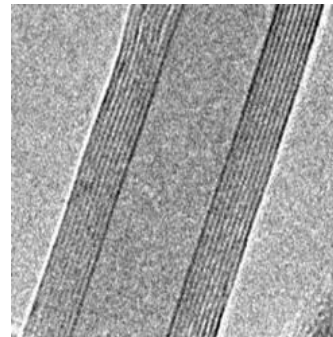
## Scanning Electron Microscopy (SEM)

- an electron beam scans the surface and the reflected (backscattered) electrons are collected
- sample must be electrically conductive
- material surface is observed
- 200,000x magnification possible



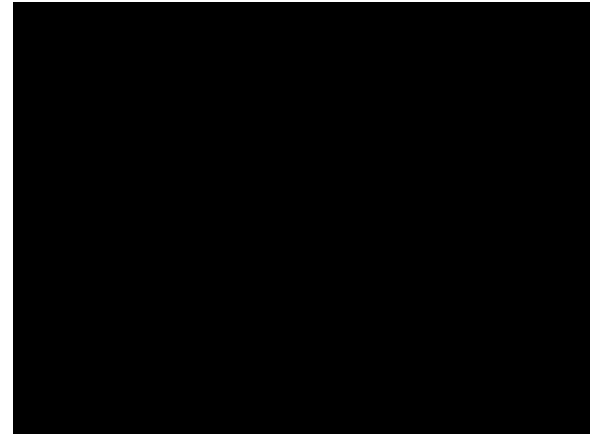
## Transmission Electron Microscopy (TEM)

- an electron beam passes through the material
- thin samples
- details of internal microstructure observed
- 1,000,000x magnification possible

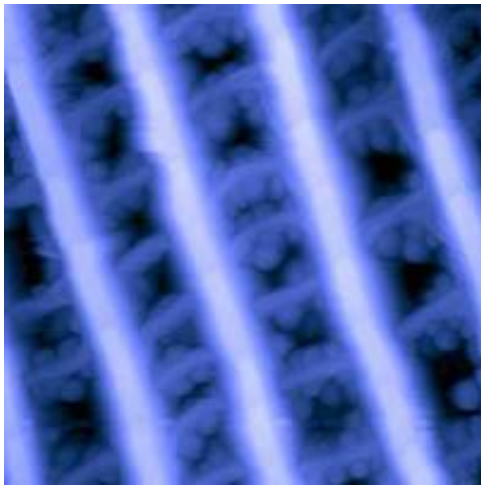


# Scanning Probe Microscopy (SPM)

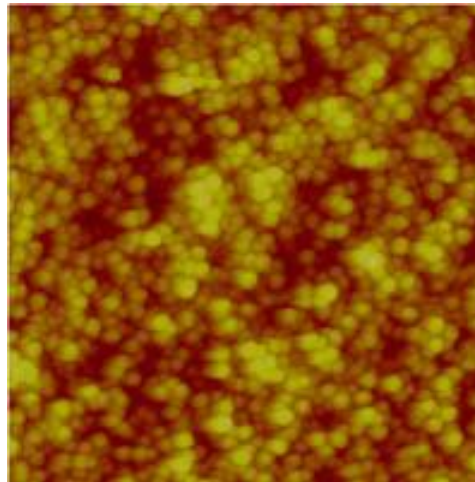
- 3D topographical map of material surface
- probe brought into close proximity of material surface
- probe rastered across the surface experiencing deflection in response to interactions with the material surface
- useful with many different types of materials



*Animation of SPM on epitaxial silicon.*



*SPM image of a butterfly wing.*

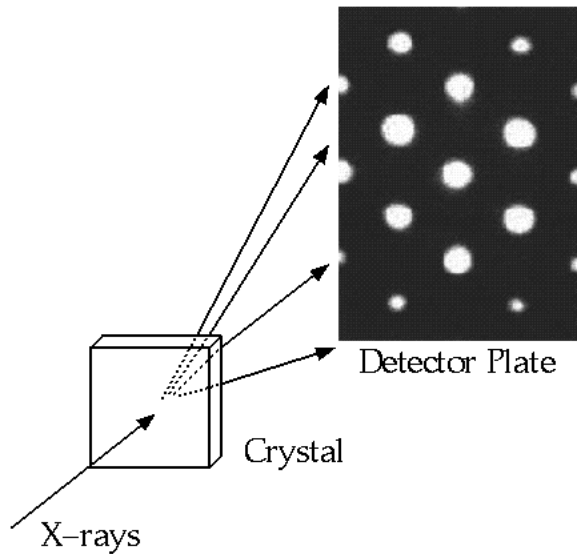


*SPM image of silica coated gold nanoparticles.*



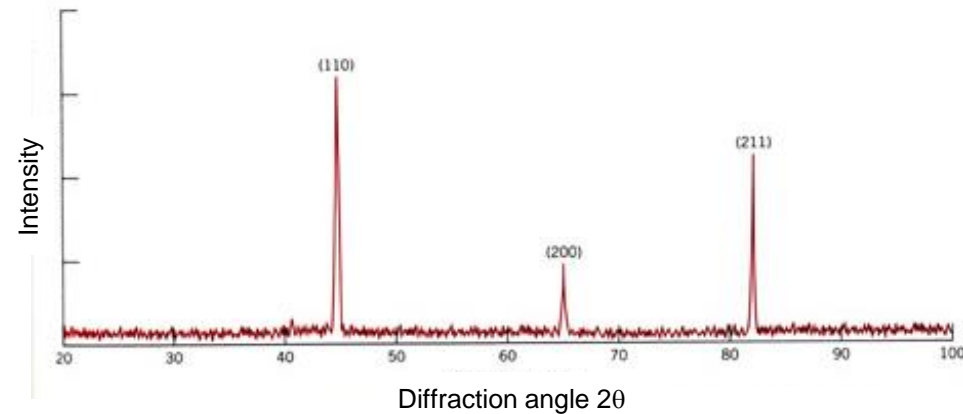
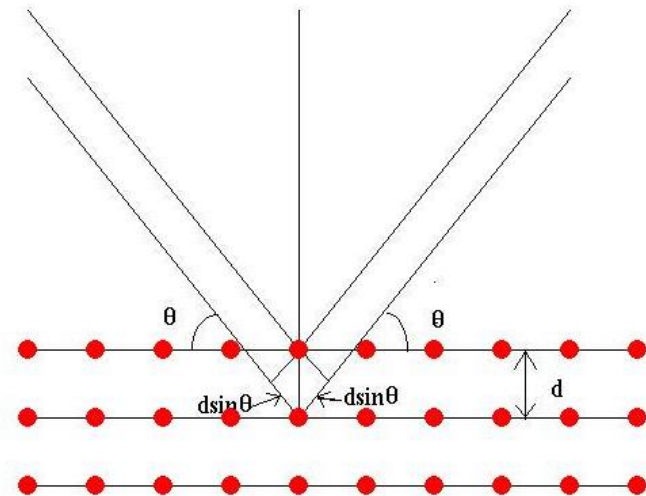
*SPM image of 70 nm photoresist lines.*

# X-ray Diffraction



- x-rays are a form of light that has high energy and short wavelength
- when x-rays strike a material a portion of them are scattered in all directions
- if the atoms in the material is crystalline or well-ordered constructive interference can occur

Bragg's Law:  $2d \sin \theta = n\lambda$



# Material Properties

- **Mechanical**
  - E.g. Modulus of elasticity, strength
- **Electrical**
  - Electrical conductivity, dielectric constant
- **Thermal**
  - Thermal expansion, thermal conductivity, heat capacity
- **Magnetic**
  - Flux density, susceptibility
- **Optical**
  - Index of refraction, reflectivity
- **Chemical (deteriorative)**
  - Oxidation, corrosion
- **Physical Properties**
  - Density,

# The Materials Selection Process

1. Pick **Application** → Determine required **Properties**

Properties: mechanical, electrical, thermal, magnetic, optical, deteriorative.

2. **Properties** → Identify candidate **Material(s)**

Material: structure, composition.

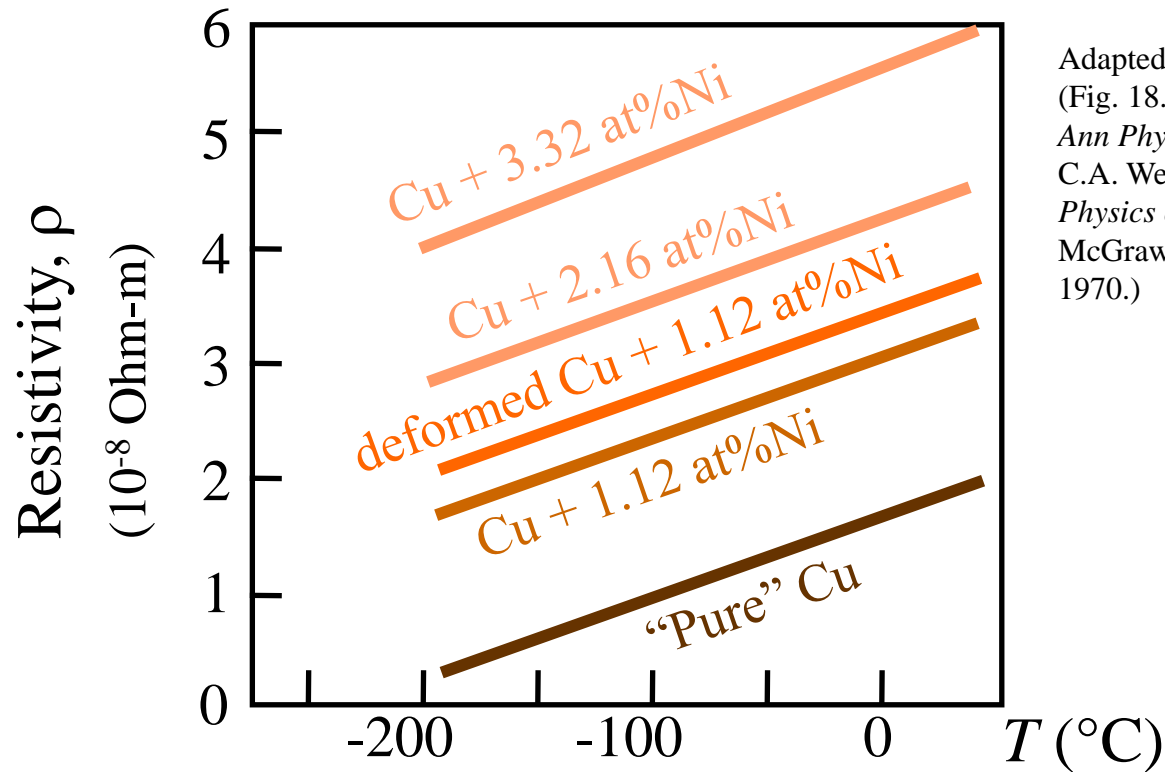
3. **Material** → Identify required **Processing**

Processing: changes *structure* and overall *shape*  
ex: casting, sintering, vapor deposition, doping  
forming, joining, annealing.



# ELECTRICAL PROPERTIES

- Electrical Resistivity of Copper:

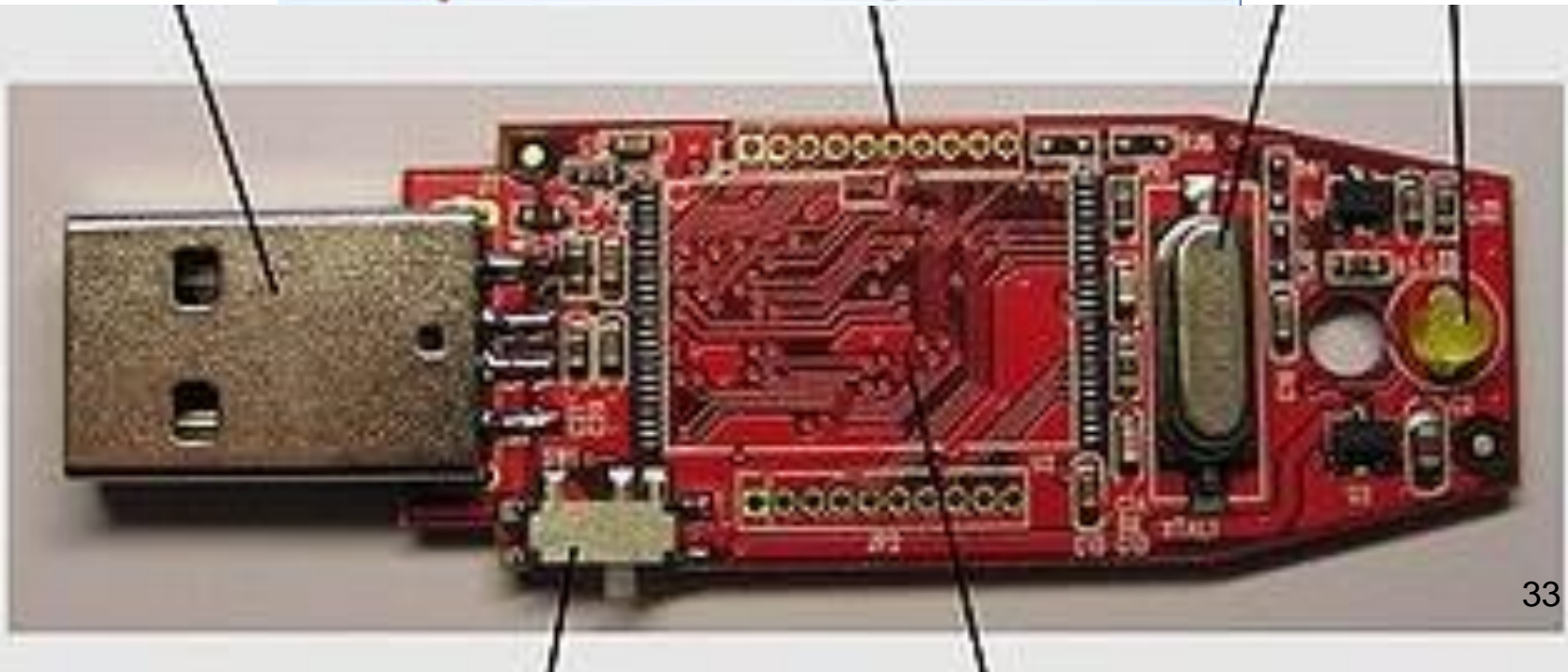
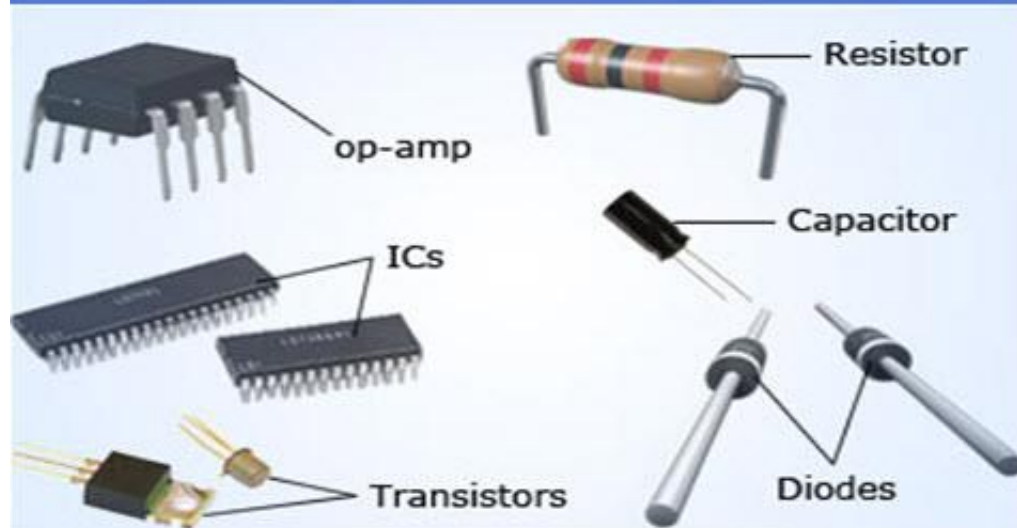


Adapted from Fig. 18.8, *Callister 7e*.  
(Fig. 18.8 adapted from: J.O. Linde, *Ann Physik* **5**, 219 (1932); and C.A. Wert and R.M. Thomson, *Physics of Solids*, 2nd edition, McGraw-Hill Company, New York, 1970.)

- Adding “impurity” atoms to Cu increases resistivity.
- Deforming Cu increases resistivity.



## Semiconductor Devices



# Types of Materials

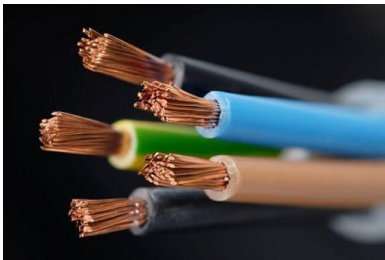
## ☐ Metals:

Metallic bonding;

Cu, Al, Ni, Fe, Au, bronze (Cu-Sn), steel (Fe-C) etc.

They are grouped as ferrous (steels) and non-ferrous (copper, magnesium, titanium and so on) metals

Properties: strong, ductile, high density, good conductors of heat and electricity (free valance electrons)



Copper electric wires



Aluminum cup



Jet engine containing mainly titanium alloys



Car body panel: low carbon steel  
Engine composed of steel and cast iron parts



Drawback:  
Corrosion of some metals, i.e. Steel, iron



- Metals
  - Pure metals
    - Iron [Fe]
    - Copper [Cu]
    - Aluminium [Al]
  - Alloys
    - Steel [Fe+C]
    - Brass [Cu+Zn]
    - Bronze [Cu+Sn]
    - Duralumin [Al+Cu]
    - Stainless steel [Fe+C+Cr+Ni],
    - Cast iron [Fe+C]

# Types of Materials

- **Polymeric (Plastic) Materials**

- **Organic giant molecules and mostly noncrystalline.**
- **Some are mixtures of crystalline and noncrystalline regions.**
- **Poor conductors of electricity and hence used as insulators.**
- **Strength and ductility vary greatly.**
- **Low densities and decomposition temperatures.**
- ***Examples :- Poly vinyl Chloride (PVC), Polyester, PTFE, Nylon***
- ***Applications :- Appliances, DVDs, Fabrics etc.***



# □Polymers: POLYMER (Greek root)

*poly = many meros = part*

Organic compounds based on C, H and other nonmetallic elements.

Large molecular structures

(e.g. Epoxy, Nylon, PVC, Polystyrene, Plastics and rubber)

Properties: weak, low density, ductile, extremely flexible, insulators.

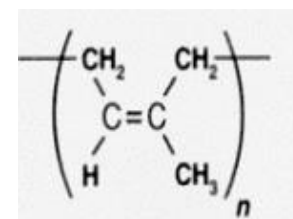
## Natural Polymers

Rubber, cotton, wool, leather, silk



## Synthetic Polymers

PP, PS, PVC, PE



atural rubber



# Recyclable plastics

## PLASTIC RESIN CODES



PETE

**Polyethylene Terephthalate**

soda bottles  
water bottles  
shampoo bottles  
mouthwash bottles  
peanut butter jars



HDPE

**High Density Polyethylene**

milk, water and juice jugs  
detergent bottles  
yogurt and margarine tubs  
grocery bags



V

**Vinyl**

clear food packaging  
shampoo bottles



LDPE

**Low Density Polyethylene**

bread bags  
frozen food bags  
squeezeable bottles (mustard, honey)



PP

**Polypropylene**

ketchup bottles  
yogurt and margarine tubs



PS

**Polystyrene**

meat trays  
egg cartons  
cups and plates



OTHER

**Other**

ketchup  
3 & 5 gallon water bottles  
some juice bottles

# Types of Materials

- **Ceramic Materials**

- Metallic and nonmetallic elements are **chemically bonded** together.
- Inorganic but can be either crystalline, noncrystalline or mixture of both.
- High hardness, strength and wear resistance.
- Very good insulator. Hence used for furnace lining for heat treating and melting metals.
- Also used in space shuttle to insulate it during exit and reentry into atmosphere.
- **Other applications** : Abrasives, construction materials, utensils etc.

➤ ***Example:-** Porcelain, Glass, Silicon nitride.*

## ☐ Ceramics:

Combinations of metals or with oxygen, nitrogen, carbon and boron (oxides, nitrides, carbides, borides)  $\text{CaO}$ ,  $\text{Al}_2\text{O}_3$ ,  $\text{BN}$ ,  $\text{SiC}$ ,  $\text{TiB}_2$

Properties: hard but very brittle, Insulators of heat and electricity, resistant to high temperature and harsh environments,



### 1-TRADITIONAL CERAMICS

Pottery, porcelain, brick, glass



Whiteware

### 2-ADVANCED CERAMICS

*Structural:* bioceramics, cutting tools, engine components, armour.

*Electrical:* Capacitors, insulators, magnets and superconductors



Brake disc



SiC engine components



SiC body armour



Cutting tools





# Types of Materials



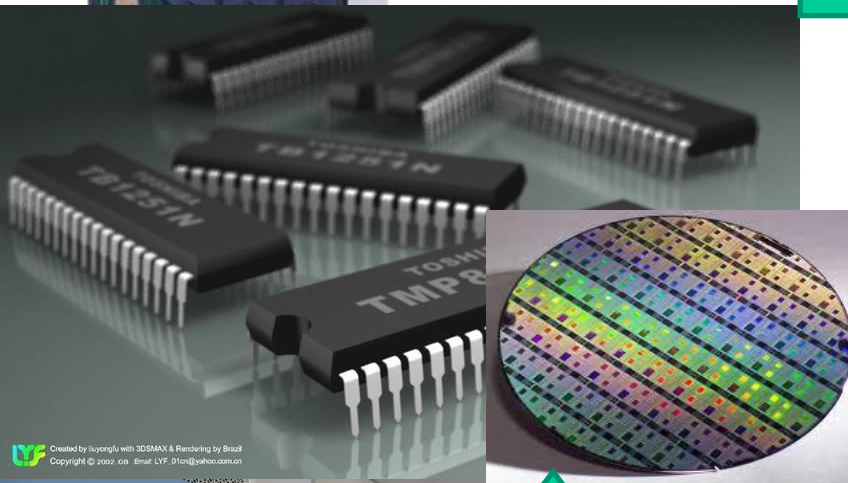
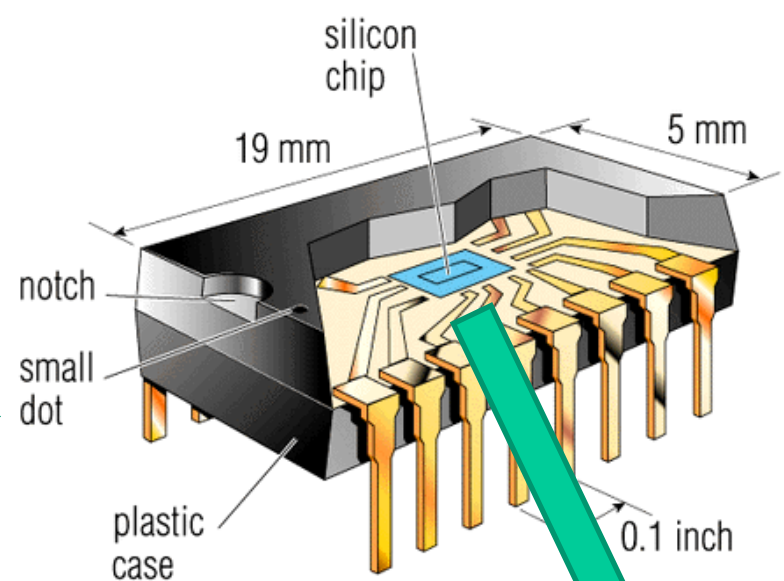
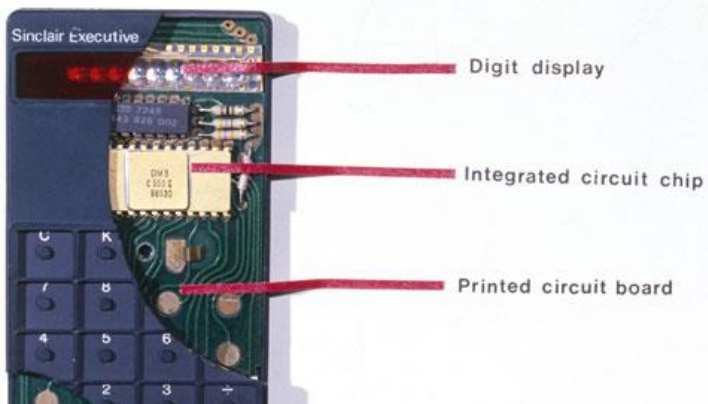
## Composite Materials

- **Mixture** of two or more materials.
- Consists of a filler material and a binding material.
- Materials only bond, will not dissolve in each other.
- Mainly two types :-
  - Fibrous: Fibers in a matrix
  - Particulate: Particles in a matrix
- Matrix can be metals, ceramic or polymer
- *Examples* :
  - Fiber Glass ( Reinforcing material in a polyester or epoxy matrix)
  - Concrete ( Gravels or steel rods reinforced in cement and sand)
- *Applications*:- Aircraft wings and engine, construction.

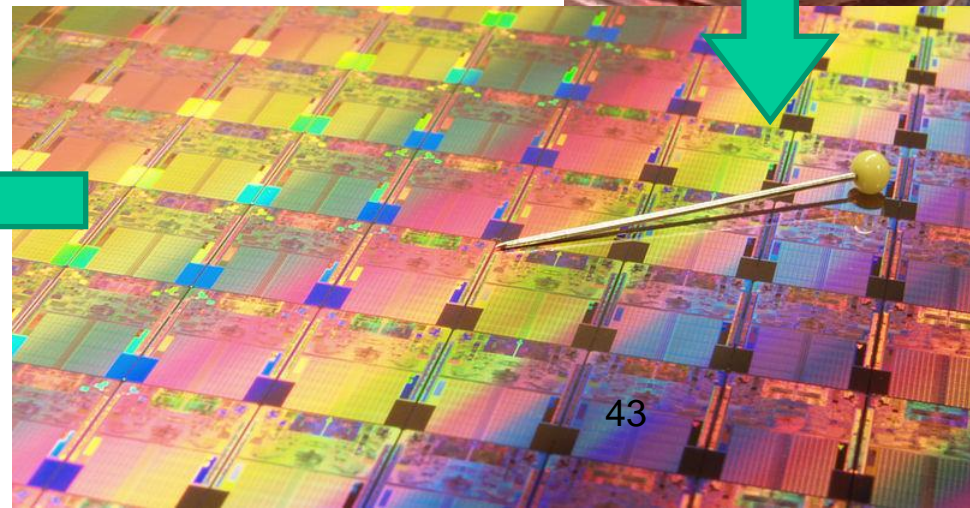
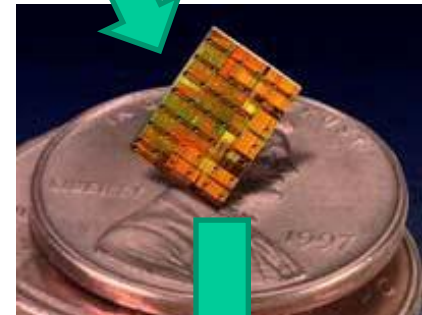
# Types of Materials

## Electronic Materials

- Not Major by volume but very important.
- Silicon is a common electronic material.
- Its electrical characteristics are changed by adding **impurities**.
- *Examples:-* Silicon chips, transistors
- *Applications :-* Computers, Integrated Circuits, Satellites etc.

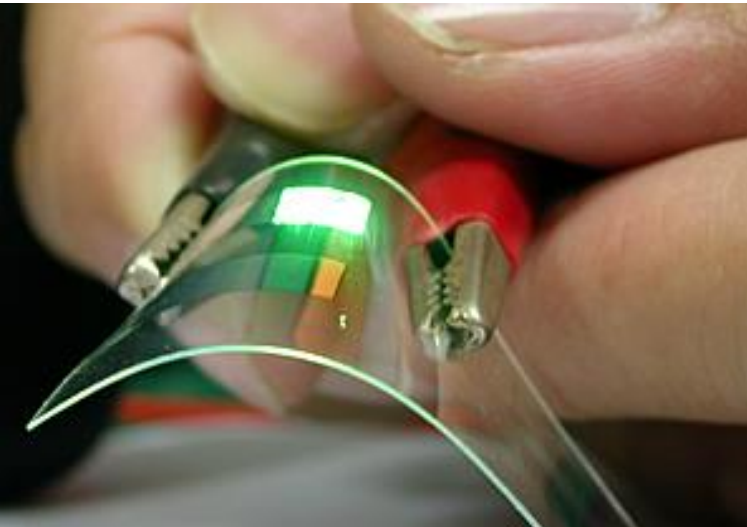


Smaller and thinner than a dime, this tiny silicon chip contains millions of transistors that work together





# SEMICONDUCTORS



**OLED  
Technology**



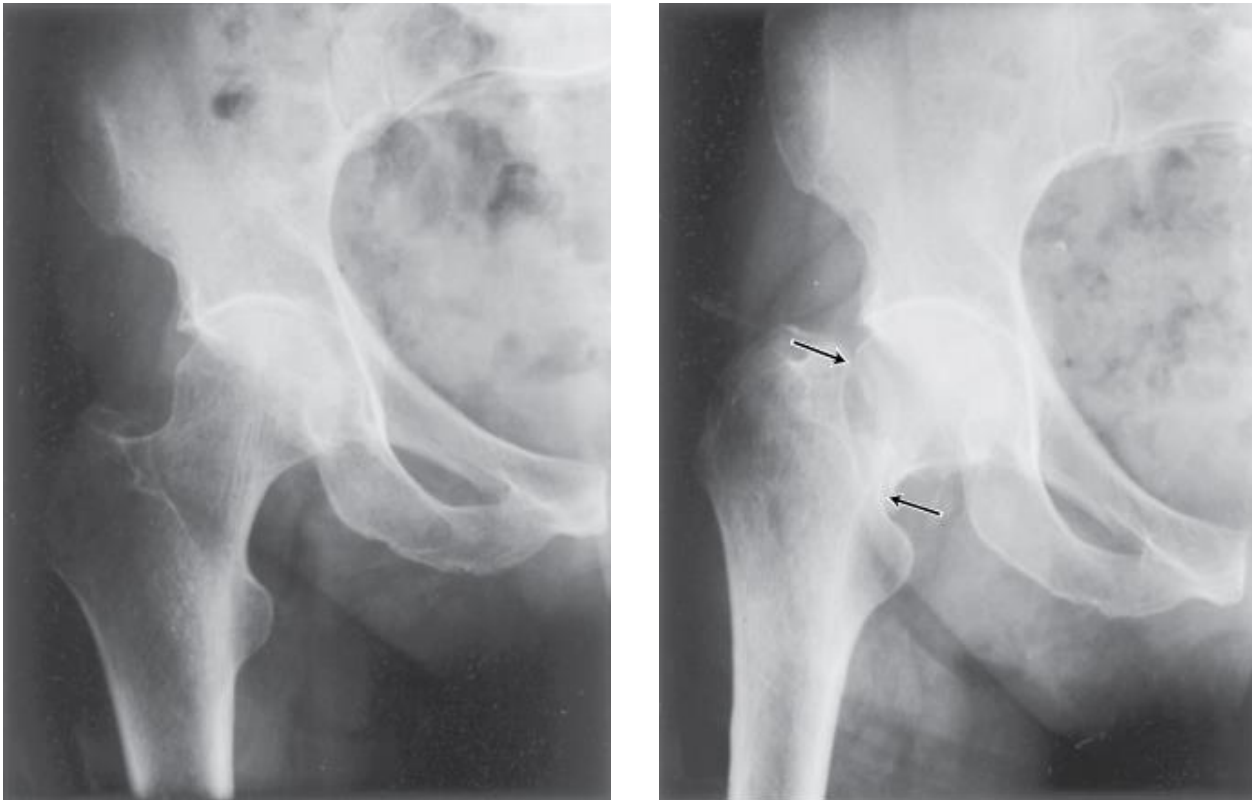
**Solar Cells**



# BIOMATERIALS

## Example – Hip Implant

- With age or certain illnesses joints deteriorate. Particularly those with large loads (such as hip).

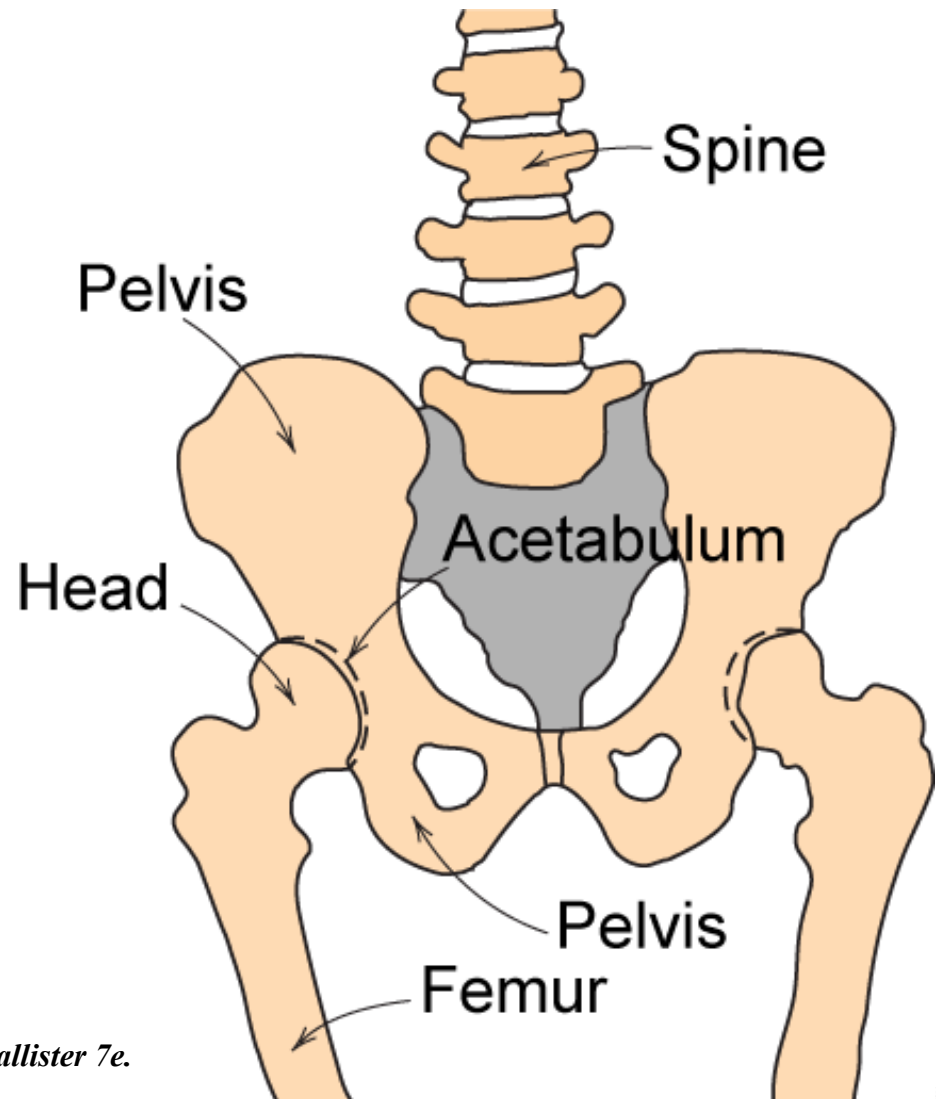


Adapted from Fig. 22.25, *Callister 7e*.



# Example – Hip Implant

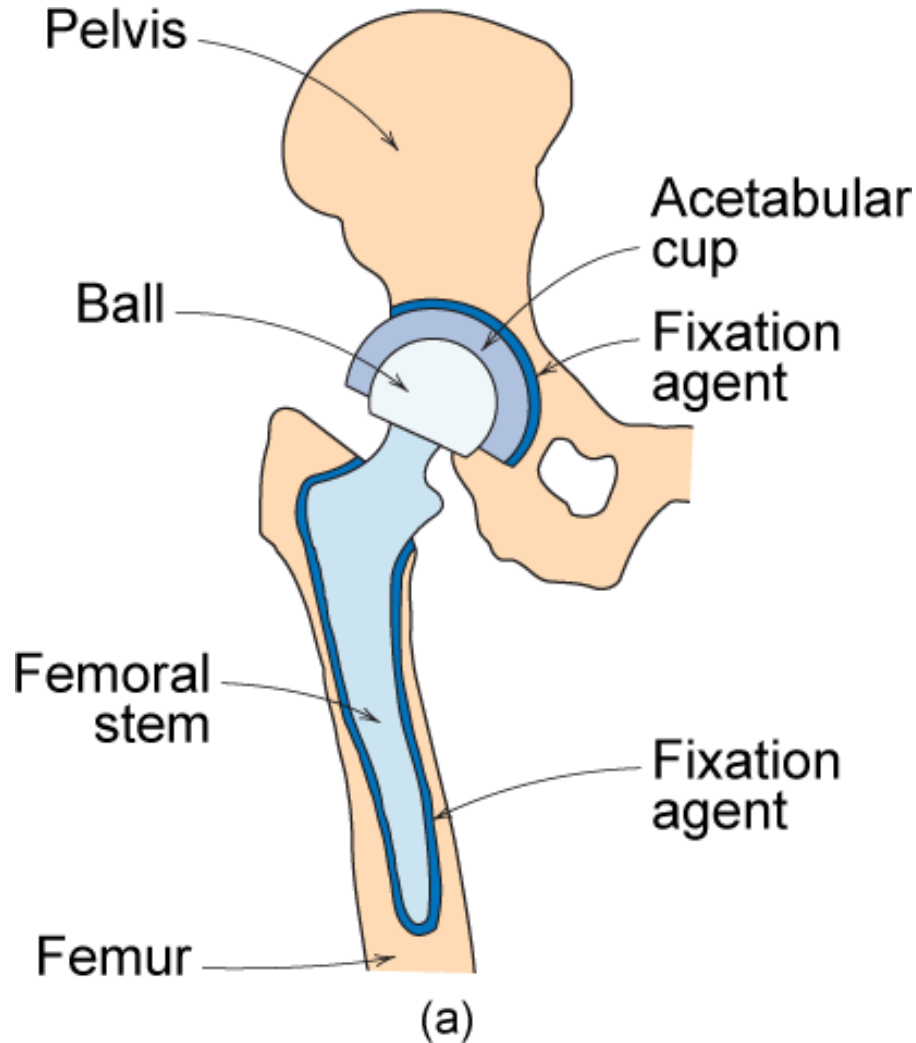
- Requirements
  - mechanical strength (many cycles)
  - good lubricity
  - biocompatibility



Adapted from Fig. 22.24, *Callister 7e*.



# Example – Hip Implant



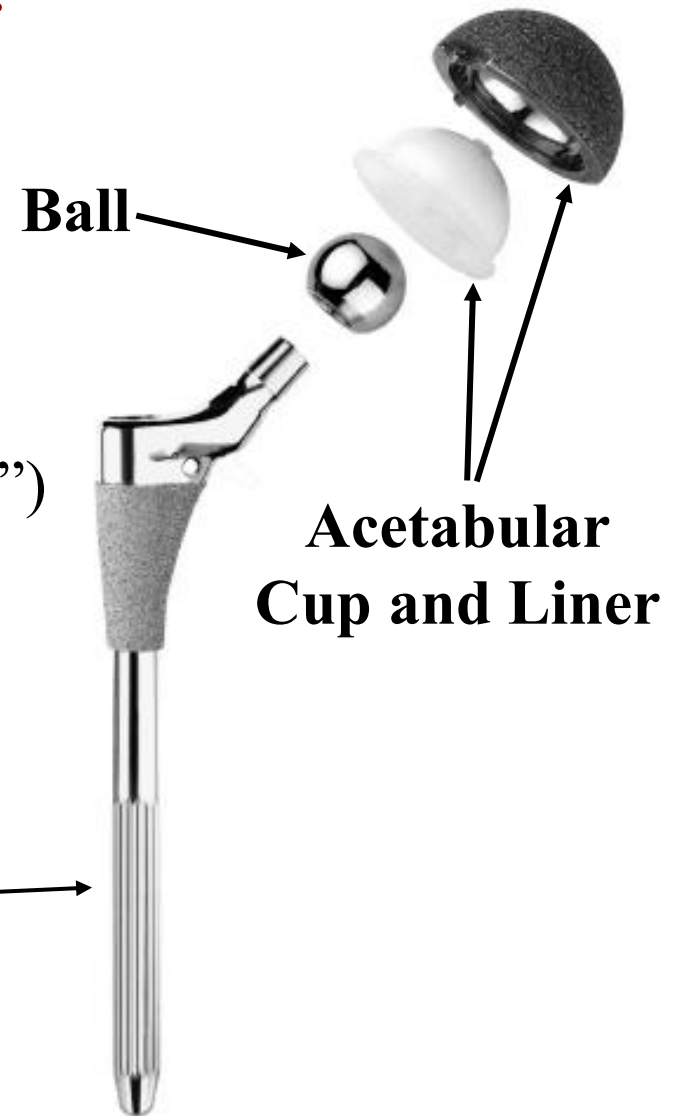
Adapted from Fig. 22.26, *Callister 7e*.



# Hip Implant

- Key problems to overcome
  - fixation agent to hold acetabular cup
  - cup lubrication material
  - femoral stem – fixing agent (“glue”)
  - must avoid any debris in cup

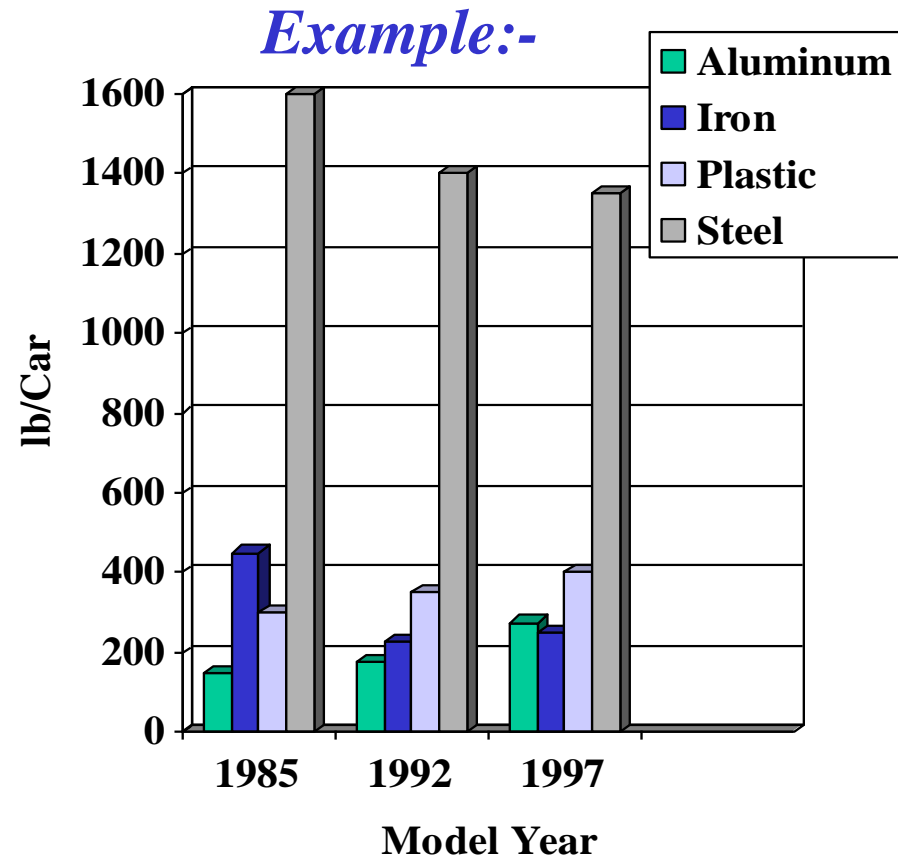
**Femoral  
Stem**



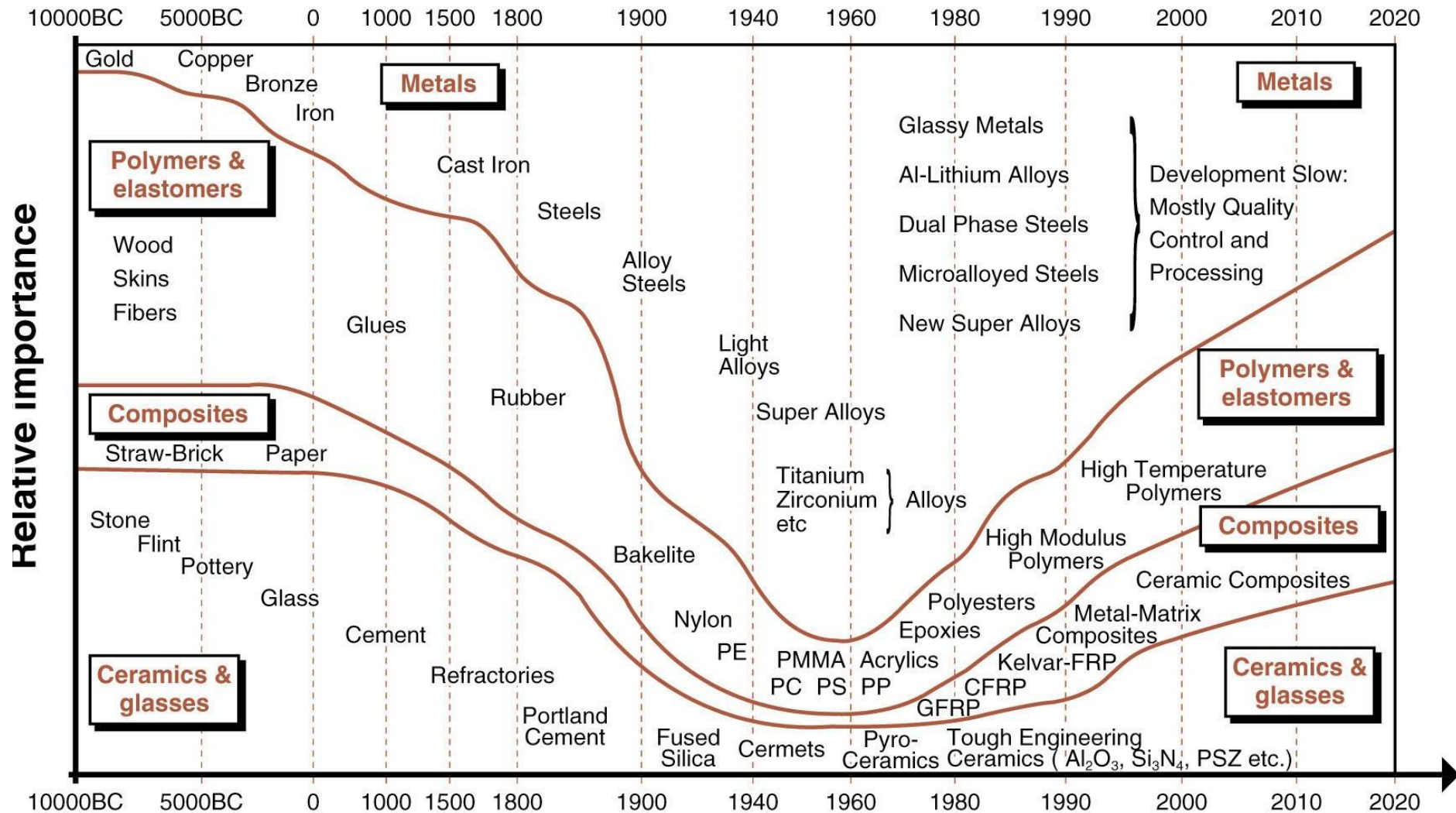
Adapted from chapter-opening photograph,  
Chapter 22, *Callister 7e*.

# Competition Among Materials

- Materials compete with each other to exist in new market
- Over a period of time usage of different materials changes depending on **cost and performance**.
- New, cheaper or better materials replace the old materials when there is a breakthrough in technology



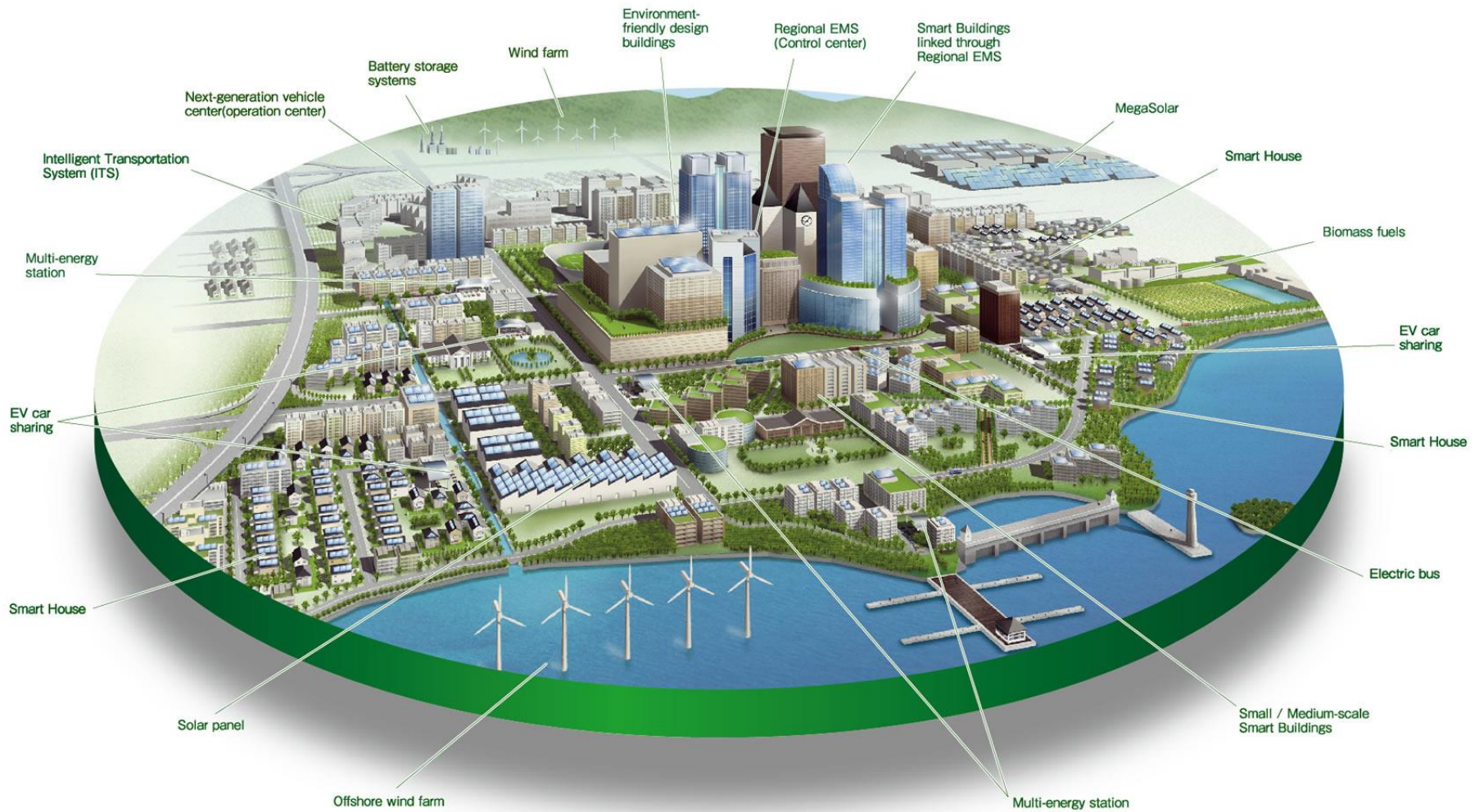
**Figure 1.14**  
**Predictions and use of**  
**materials in US automobiles.**



# Advanced Materials

- **High tech Materials**-(intricate and sophisticated principles)
  - Materials used for laser Technology
  - Integrated circuits
  - Magnetic information storage
  - Liquid crystal displays
  - Fiber optics
  - Aerospace applications
- **Smart Materials**-(Material properties change with the environment in a predetermined manner) used in sensors and actuators
  - Shape memory alloys
  - Piezoelectric ceramics
  - Magnetostrictive materials
  - Electrorheological/magnetorheological fluids
- **NanoTechnology**

# Sustainable city concept





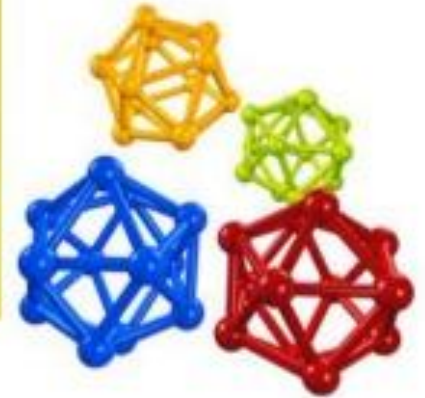
# SMART CITY





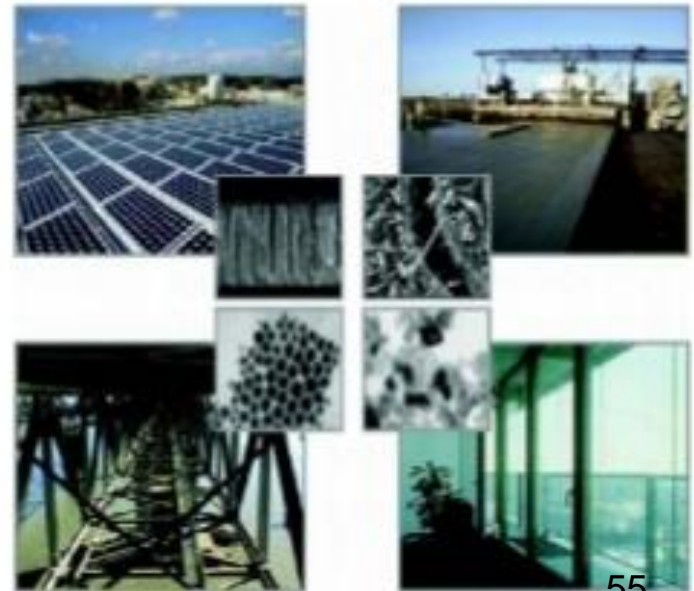


# Applications of Nano-technology









Nanotechnology is widely used in construction material as:

- In Concrete
- In Steel
- In Wood
- In Glass
- In Coating
- In photovoltaic



Existing applications

Possible future applications

Application	Functionalities	Car body shell exterior	Car body	Interior	Chassis and tyres	Electrics and electronics	Engine and drive train
Effect							
Mechanical functionalities	Hardness, friction, tribological properties, breaking resistance	Nano varnish			Carbon black in tyres		Low-friction aggregate components
		Polymer glazing	Nanosteel		Nanosteel		
Geometric effects	Large surface-to-volume ratio, Poresize			Nano filter		Super caps	
			Gecko effect	Gecko effect		Fuel cell	
Electronic/magnetic functionalities	Size dependent electric and magnetic properties		Gluing on command			GMR sensors	Piezo injectors
					Switchable materials (rheology)	Solar cells	
Optical functionalities	Colour, fluorescence, transparency	Ultra-thin layers		Anti-glare coatings			
		Electro chromatic layers					
Chemical functionalities	Reactivity, selectivity, surface properties	Care and sealing systems	Forming of high strength steel	Dirt protection			Catalysts
			Corrosion protection	Fragrance in the cabin			Fuel additives





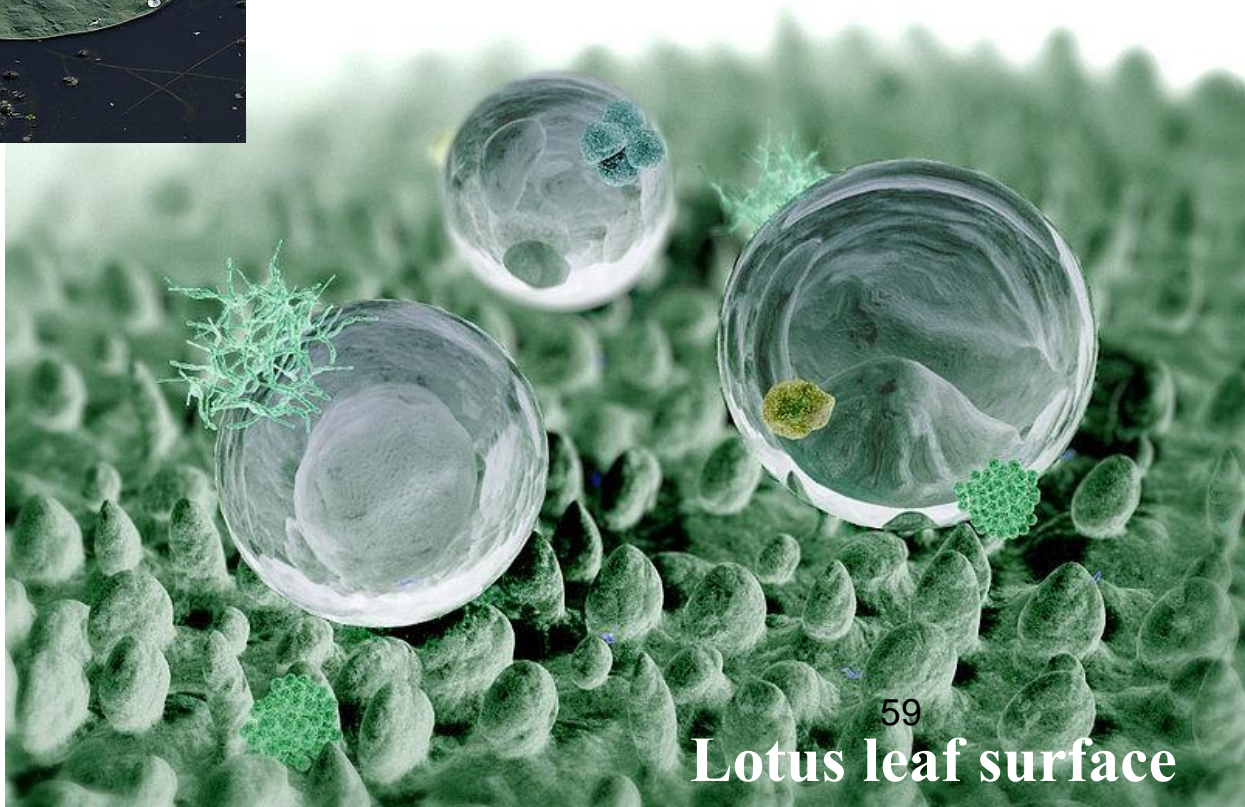




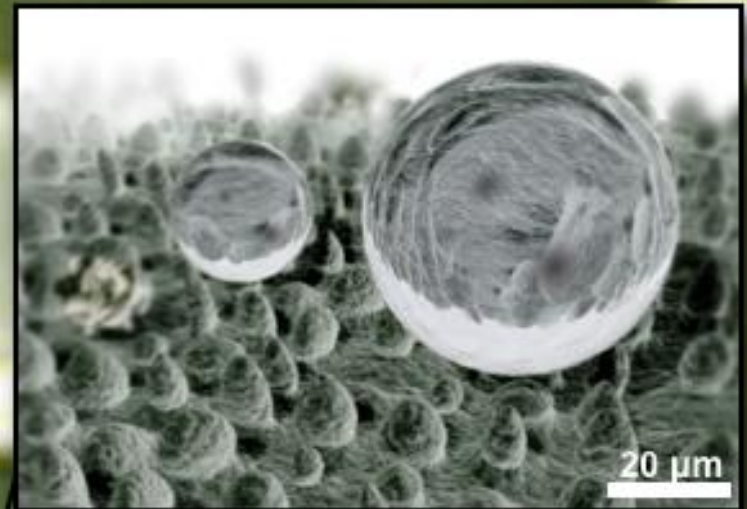
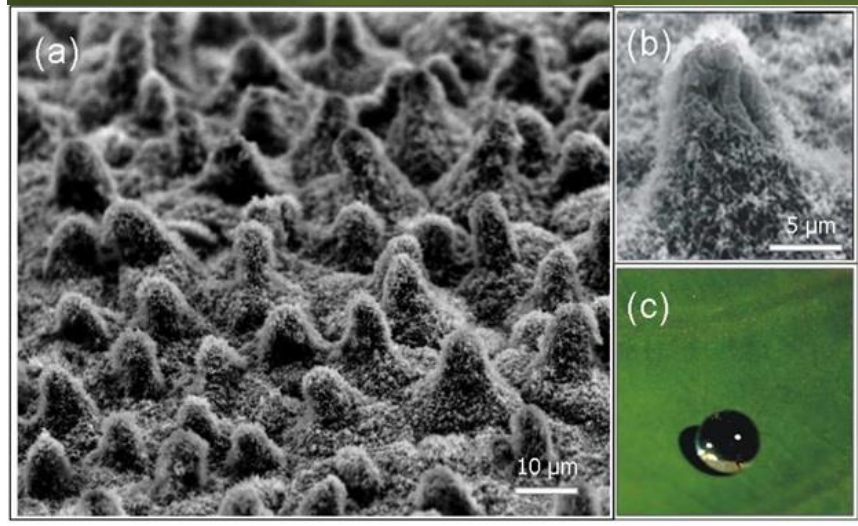
# BIOMIMETICS



Some paints and roof tiles have been engineered to be self-cleaning by copying the mechanism from the **lotus**



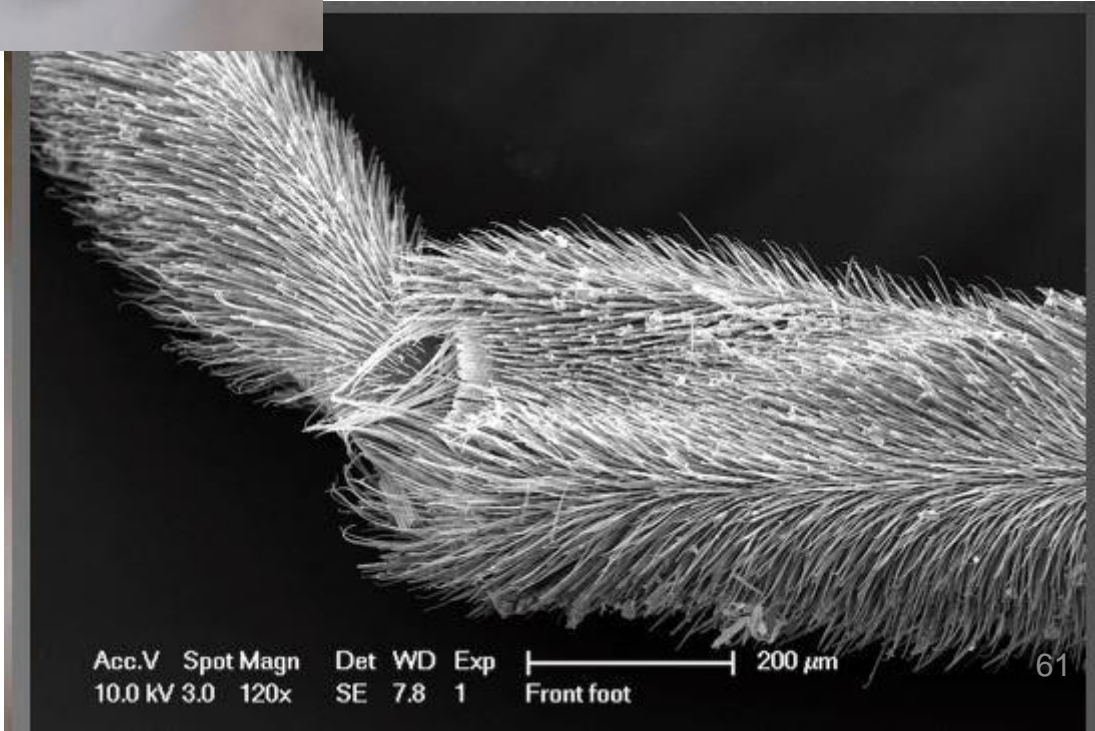




**Lotus leaves** are anti-wetting because of a reduced **contact area** from **surface texture** combined with low surface energy **chemical functionality**



Water strider



Acc.V Spot Magn Det WD Exp | 200  $\mu$ m  
10.0 kV 3.0 120x SE 7.8 1 Front foot

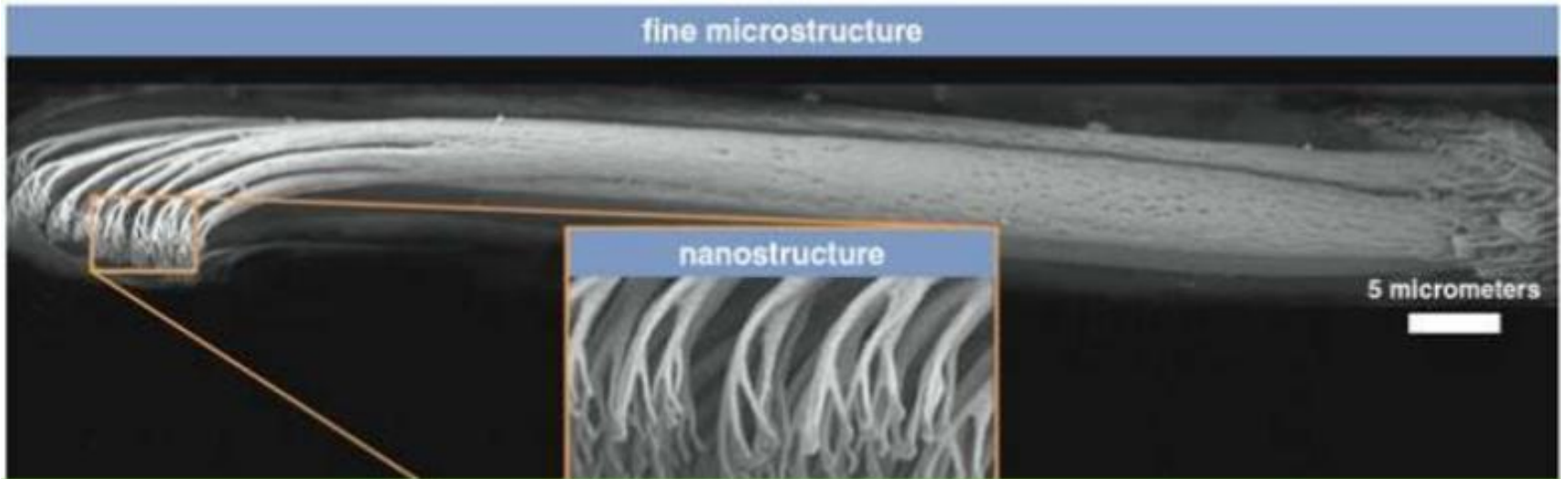
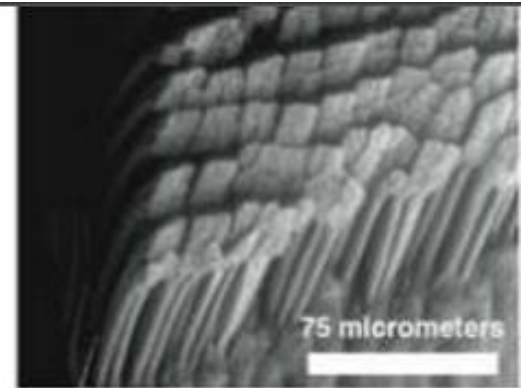
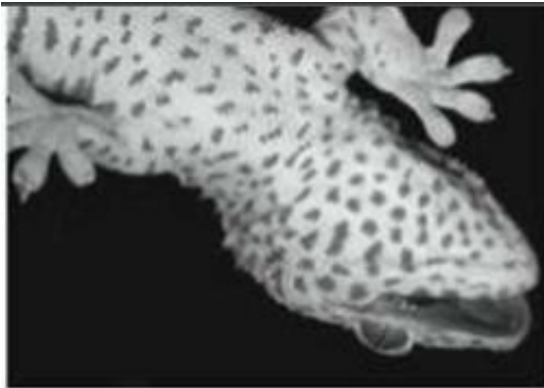


A force similar to surface tension,  
called van der Waals, lets geckos walk on walls!



### How is this nano?

Different physical forces dominate when things get very, very small. For example, gravity is very obvious at human size, but it's hardly noticeable to nano-sized things like water molecules. Other forces (like surface tension) are much more important.



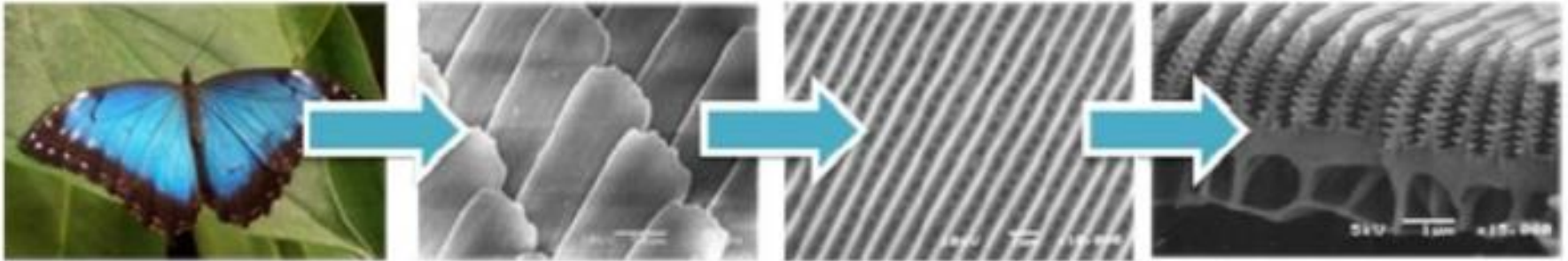


## Case of Butterfly

The wings of Butterflies often display extra ordinary colours which are a consequence of the wings' surface and its interaction with light.

It also exhibits the phenomenon of IRIDESCENCE.

# SEM Analysis of Wings



- Shows even more intricate structure called SETAE : looks like fir trees
- About 400 nm long, responsible for producing constructive interference in blue wavelengths which generates strong blue colour.



# Toucan's Beak: Strong and light

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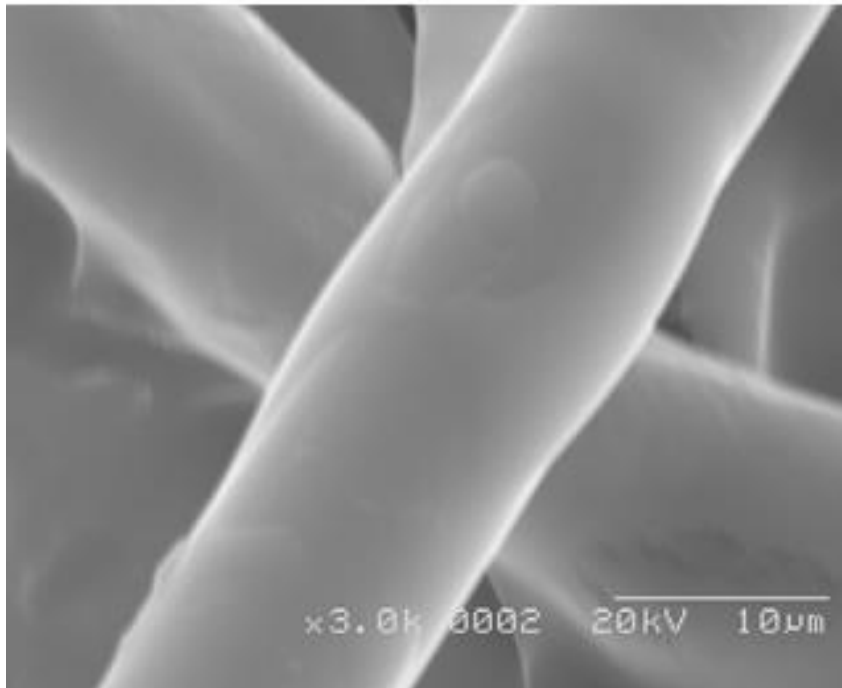
Despite its large size (a third of the length of the bird) and considerable strength, the toucan beak comprises only one twentieth the bird's mass.

While the large strong beak is useful in foraging, defense and attracting mates, its low density is essential for the toucan to retain its ability to fly.



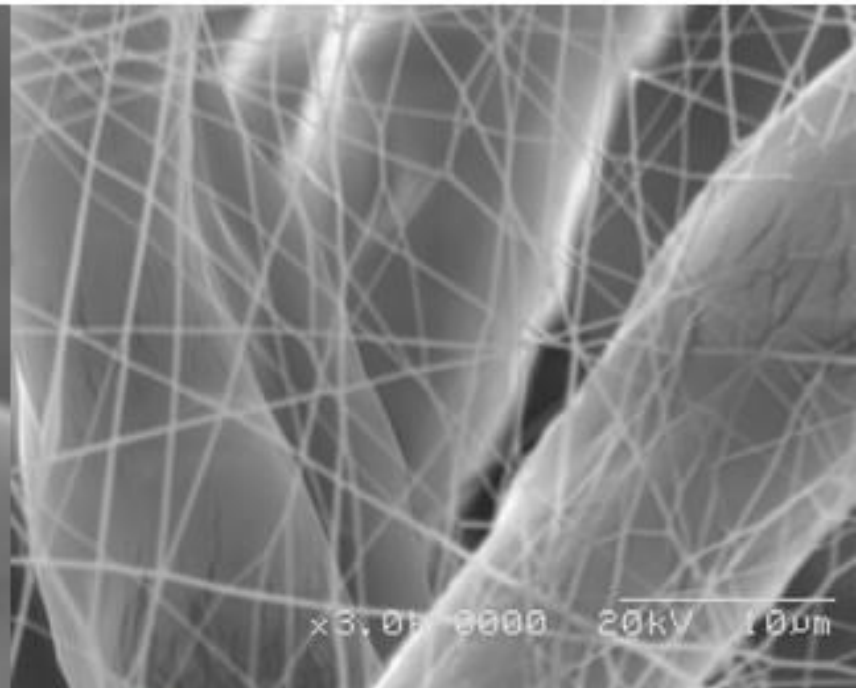
# cobweb

Our general filter media



3.0 K times (SEM)

NANOWOVEN



3.0 K times (SEM)



Cobweb-like nano fibers can be observed



## Biomimetic material



- ☐ Adhesives
- ☐ High tensile strength fibre
- ☐ Dirt and water resistant paint
- ☐ IMOD Display technology
- ☐ Reduced drag suits for athletes
- ☐ Structural elements
- ☐ Aerospace and automotive applications

## Inspired from



- ☐ Gecko's feet
- ☐ Spider silk
- ☐ Lotus leaf
- ☐ Butterfly wings
- ☐ Shark skin
- ☐ Wood, ligaments and bone
- ☐ Toucan's beak

# **SOME CURRENT APPLICATIONS OF NANOTECHNOLOGY**



# SOLAR CELLS

Nanotechnology enhancements provide:

- **Improved efficiencies:** novel nanomaterials can harness more of the sun's energy
- **Lower costs:** some novel nanomaterials can be made cheaper than alternatives
- **Flexibility:** thin film flexible polymers can be manipulated to generate electricity from the sun's energy



# COMPUTING

Nanotechnology enhancements provide:

- **Faster processing speeds:**  
miniaturization allows more transistors to be packed on a computer chip
- **More memory:** nanosized features on memory chips allow more information to be stored
- **Thermal management solutions for electronics:** novel carbon-based nanomaterials carry away heat generated by sensitive electronics



# CLOTHING

Nanotechnology enhancements provide:

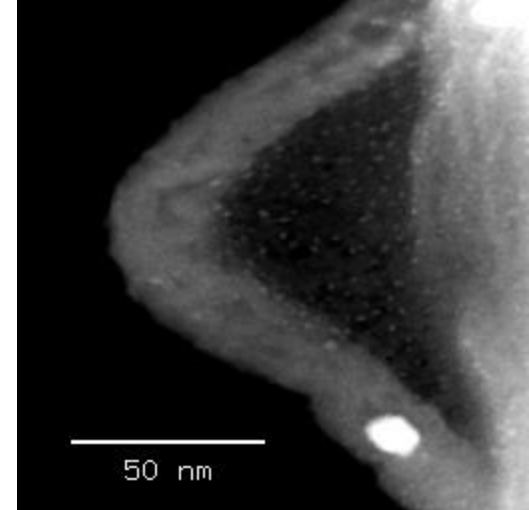
- **Anti-odor properties:** silver nanoparticles embedded in textiles kill odor causing bacteria
- **Stain-resistance:** nanofiber coatings on textiles stop liquids from penetrating
- **Moisture control:** novel nanomaterials on fabrics absorb perspiration and wick it away
- **UV protection:** titanium nanoparticles embedded in textiles inhibit UV rays from penetrating through fabric



# BATTERIES

**Nanotechnology enhancements provide:**

- **Higher energy storage capacity and quicker recharge: nanoparticles or nanotubes on electrodes provide high surface area and allow more current to flow**
- **Longer life: nanoparticles on electrodes prevent electrolytes from degrading so batteries can be recharged over and over**
- **A safer alternative: novel nano-enhanced electrodes can be less flammable, costly and toxic than conventional electrodes**





# SPORTING GOODS AND EQUIPMENT

**Nanotechnology enhancements provide:**

- **Increased strength of materials:**  
novel carbon nanofiber or nanotube-based nanocomposites give the player a stronger swing
- **Lighter weight materials:**  
nanocomposites are typically lighter weight than their macroscale counterparts



# CARS

**Nanotechnology enhancements provide:**

- **Increased strength of materials:**  
novel carbon nanofiber or nanotube nanocomposites are used in car bumpers, cargo liners and as step-assists for vans
- **Lighter weight materials:**  
lightweight nanocomposites mean less fuel is used to make the car go
- **Control of surface characteristics:** nanoscale thin films can be applied for optical control of glass, water repellency of windshields and to repair of nicks/scratches



# FOOD AND BEVERAGE

**Nanotechnology enhancements provide:**

- **Better, more environmentally friendly adhesives for fast food containers**
- **Anti-bacterial properties: Nano silver coatings on kitchen tools and counter-tops kill bacteria/microbes**
- **Improved barrier properties for carbonated beverages or packaged foods: nanocomposites slow down the flow of gas or water vapor across the container, increasing shelf life**



# THE ENVIRONMENT

**Nanotechnology enhancements provide:**

- **Improved ability to capture groundwater contaminants: nanoparticles with high surface area are injected into groundwater to bond with contaminants**
- **Replacements for toxic materials**



# BODY ARMOR

Nanotechnology enhancements will provide...

- Stronger materials for better protection: nanocomposites that provide unparalleled strength and impact resistance
- Flexible materials for more form-fitting wearability: nanoparticle-based materials that act like “liquid armor”
- Lighter weight materials: nanomaterials typically weigh less than their macroscale counterparts
- Dynamic control: nanofibers that can be flexed as necessary to provide CPR to soldiers or stiffen to furnish additional protection in the face of danger

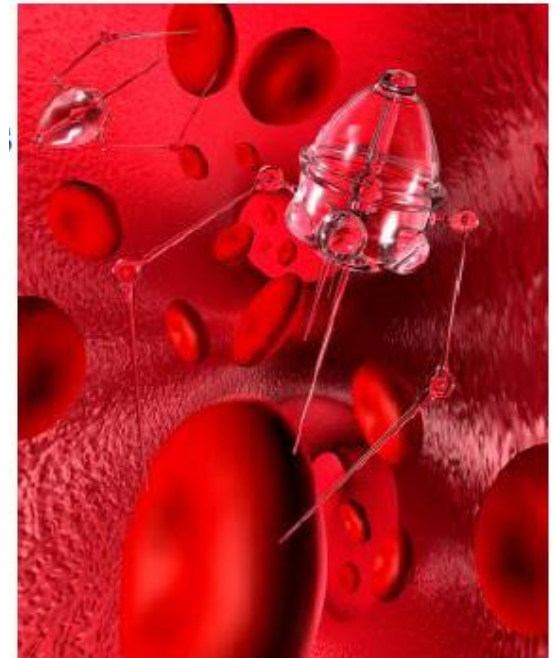




# DRUG DELIVERY

Nanotechnology enhancements will provide:

- **New vehicles for delivery:** nanoparticles such as buckyballs or other cage-like structures that carry drugs through the body
- **Targeted delivery:** nano vehicles that deliver drugs to specific locations in body
- **Time release:** nanostructured material that store medicine in nanosized pockets that release small amounts of drugs over time



# CANCER

Nanotechnology enhancements will provide:

- **Earlier detection: specialized nanoparticles that target cancer cells only – these nanoparticles can be easily imaged to find small tumors**
- **Improved treatments: infrared light that shines on the body is absorbed by the specialized nanoparticles in the cancer cells only, leading to an increased localized temperature that selectively kills the cancer cells but leaves normal cells unharmed**

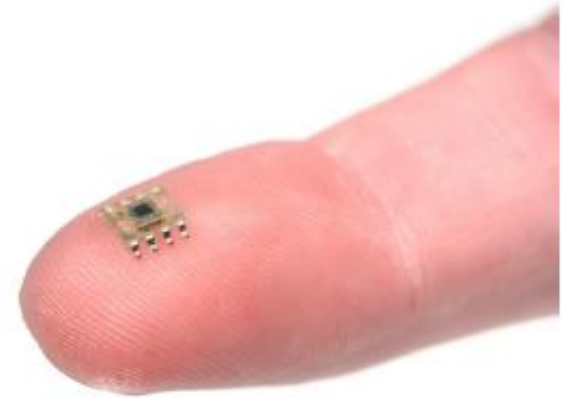


*Image of a cancer cell*

# SENSORS

**Nanotechnology enhancements will provide:**

- **Higher sensitivity:** high surface area of nanostructures that allows for easier detection of chemicals, biological toxins, radiation, disease, etc.
- **Miniaturization:** nanoscale fabrication methods that can be used to make smaller sensors that can be hidden and integrated into various objects



# NEXT GENERATION COMPUTING

**Nanotechnology enhancements will provide:**

- **The ability to control atomic scale phenomena: quantum or molecular phenomena that can be used to represent data**
- **Faster processing speeds**
- **Lighter weight and miniaturized computers**
- **Increased memory**
- **Lower energy consumption**



# NANOROBOTICS

**Nanotechnology enhancements will provide:**

- **Miniaturized fabrication of complex nanoscale systems: nanorobots that propel through the body and detect/ cure disease or clandestinely enter enemy territory for a specific task**
- **Manipulation of tools at very small scales: nanorobots that help doctors perform sensitive surgeries**





# WATER PURIFICATION

**Nanotechnology enhancements will provide:**

- **Easier contamination removal:**  
filters made of nanofibers that can remove small contaminants
- **Improved desalination methods:**  
nanoparticle or nanotube membranes that allow only pure water to pass through
- **Lower costs**
- **Lower energy use**



# MORE ENERGY/ENVIRONMENT APPLICATIONS...

**Nanotechnology enhancements will provide:**

- Improvements to solar cells
- Improvements to batteries
- Improvements to fuel cells
- Improvements to hydrogen storage
- CO<sub>2</sub> emission reduction: nanomaterials that do a better job removing CO<sub>2</sub> from power plant exhaust
- Stronger, more efficient power transmission cables: synthesized with nanomaterials



## Future Trends

- **Smart Materials** : Change their properties by sensing external stimulus.
  - **Shape memory alloys**: Strained material reverts back to its original shape above a critical temperature.
    - Used in heart valves and to expand arteries.
  - **Piezoelectric materials**: Produce electric field when exposed to force and vice versa.
    - Used in actuators and vibration reducers.

# MEMS and Nanomaterials

- **MEMS: Microelectromechanical systems.**
  - Miniature devices
  - Micro-pumps, sensors
- **Nanomaterials:** Characteristic length  $< 100$  nm
  - Examples: ceramics powder and grain size  $< 100$  nm
  - Nanomaterials are harder and stronger than bulk materials.
  - Have **biocompatible** characteristics ( as in Zirconia)
  - Transistors and diodes are developed on a nanowire.



# Modern Materials Needs

- In the field of Nuclear Energy
  - Environmental issues
  - Disposal of radioactive waste
- Transportation
  - Energy and environmental issues
- Energy
  - Advanced Solar cells
  - Light emitting diodes - LEDs
- Pollution control
  - Catalytic converter
- Recycling



## Case Study – Material Selection

- **Problem:** Select suitable material for bicycle frame and fork.

